

1992B. Answer the following two periodic table questions.

Classifying the Periodic Table Grid

By using a color-coded periodic table, answer the next:

Why?

In addition, the periodic table has been divided into groups. The organization of the table appears to be according to the type of ions that are produced by the elements. For the first row, the elements are grouped into 2 categories: metals and nonmetals. The periodic table is organized by properties, both chemical and physical. The organization of the table shows the order of the groups of elements in the periodic table. The elements are grouped into metals, nonmetals, and metalloids. The periodic table is organized into groups and periods. The elements are grouped into metals, nonmetals, and metalloids. The periodic table is organized into groups and periods.

Blank Periodic Table

6. Classify the elements in the same group from least reactive:

Group 1: Li, Na, K, Rb, Cs, Fr

Group 2: Be, Mg, Ca, Sr, Ba, Ra

Group 17: F, Cl, Br, I, At

Group 18: He, Ne, Ar, Kr, Xe, Rn

Group 19: H, Tl, Bi, Po, At, Fr

Group 20: Be, Mg, Ca, Sr, Ba, Ra

Group 21: Sc, Y, La, Ac

Group 22: Ti, Zr, Hf, Rf

Group 23: Scandium, Yttrium, Lanthanum, Actinium

Group 24: Scandium, Yttrium, Lanthanum, Actinium

Cracking The Periodic Table Code Pogil Answer Key

Mark Steyvers



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