

**Chemistry 2A**  
**Lab 11: Molecular Geometry**

Draw Lewis Diagrams for the following molecules/ions, determine the number of bonding electron pairs, number of lone pairs, the molecular geometry, symmetry and polarity:

Molecule or Ion	Lewis Diagram	Central Atom Bonding Pairs	Central Atom Lone Pairs	Molecular Geometry	Bond Polarity	Symmetry	Molecular Polarity
1. ICl		1	6	Linear			
2. SiH <sub>4</sub>		4	0	Tetrahedral			
3. PH <sub>3</sub>		3	1	Tetrahedral			
4. H <sub>2</sub> S		2	2	Tetrahedral			

# Lab Molecular Geometry Team Chemistry

**Karin Nielsen-Saines**



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