

HOLT MODERN CHEMISTRY -- CHAPTER 2 REVIEW

The following pages contain the bulk (but not all) of the information for the chapter 2 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and **NOTHING** else, you should at least be able to **PASS** the test. ***** Test items will be recall, examples, and/or application of this content. *****

OUTCOMES

- 2.1: Describe the scientific method process. *(T & R)*
- 2.2: Measure length, volume, mass, and time using metric units. ($1 \text{ mL} = 1 \text{ cm}^3$ & $1 \text{ L} = 1 \text{ dm}^3$) *(T & R)*
- 2.2: Use volume and mass to determine density; compare relative densities of solids, liquids, gases; compare relative densities of pure liquids and solutions. *(F & PK)*
- 2.2: Use density to identify substances in activities involving collection of data and precision and accuracy in measurements. *(F & PK)*
- 2.3: Distinguish between precision and accuracy. *(T & R)*
- 2.3: Determine and perform calculations using significant figures. *(F & PK)*
- 2.3: Convert from numbers to scientific notation and reverse. *(F & PK)*

2.1 SCIENTIFIC METHOD

- **Vocabulary**
 - **scientific method** -- a series of steps followed to solve problems, including collecting data, formulating a hypothesis, testing the hypothesis, and stating conclusions
 - **system** -- a set of particles or interacting components considered to be a distinct physical entity for the purpose of study
 - **hypothesis** -- an explanation that is based on prior scientific research or observations and that can be tested
 - **model** -- a pattern, plan, representation, or description designed to show the structure or workings of an object, system, or concept
 - **theory** -- an explanation for some phenomenon that is based on observation, experimentation, and reasoning
- **Chapter Highlights**
 - The scientific method is a logical approach to solving problems that lend themselves to investigation.
 - A hypothesis is a testable statement that serves as the basis for predictions and further experiments. It is usually written as an if/then statement.
 - A theory is a broad generalization that explains a body of known facts or phenomena.
 - Communication is a very important aspect of all parts of the scientific method, including successful and/or unsuccessful results.

2.2 UNITS OF MEASUREMENT

- **Vocabulary**
 - **quantity** -- something that has magnitude, size, or amount
 - **SI** -- Le Systeme International d'Unites, or the International System of Units, which is the measurement system that is accepted worldwide
 - **weight** -- a measure of the gravitational force exerted on an object; its value can change with the location of the object in the universe
 - **derived unit** -- a unit of measure that is a combination of other measurements
 - **volume** -- a measure of the size of a body or region in three-dimensional space
 - **density** -- the ratio of the mass of a substance to the volume of the substance; often expressed as grams per cubic centimeter for solids and liquids and as grams per liter for gases
 - **conversion factor** -- a ratio that is derived from the equality of two different units and that can be used to convert from one unit to the other
 - **dimensional analysis** -- a mathematical technique for studying dimensions of physical quantities

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