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Student Exploration: pH Analysis

Vocabulary: acid, acidic, alkaline, base, indicator, neutral, pH

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- Acids** are substances that produce hydrogen ions (H^+) when dissolved in water. Lemon juice is an example of an acid.
 - What does lemon juice taste like? It tastes sharp and tangy.
 - What does it feel like if lemon juice gets in your eye? It would sting and burn your eyes.
- Bases** are substances that produce hydroxide ions (OH^-) when dissolved in water. Hand soap is an example of a base.
 - What does soap feel like? It is slippery and smooth.
 - What does soap taste like? It tastes bitter.
 - What does it feel like if soap gets in your eye? It stings and burns your eyes.

Gizmo Warm-up

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess H^+ ions there are in a solution. The pH scale runs from 0 to 14, with 0 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered **neutral**.

The *pH Analysis* Gizmo allows you to find the pH of a variety of liquids. In the Gizmo, check that the **Substance in the tube** is **Ammonia**, and click **Test**. Wait until the animation is finished.



- Indicators** change color in acids or bases. What is the color of the pH paper? Dark Blue
- Compare the paper to the **pH color chart**. What is the pH of ammonia? 11
- Is ammonia acidic or alkaline? Alkaline

Student Exploration Ph Analysis Answer Key

Michael Hoelscher



Student Exploration Ph Analysis Answer Key:

The book delves into Student Exploration Ph Analysis Answer Key. Student Exploration Ph Analysis Answer Key is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Student Exploration Ph Analysis Answer Key, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Student Exploration Ph Analysis Answer Key
 - Chapter 2: Essential Elements of Student Exploration Ph Analysis Answer Key
 - Chapter 3: Student Exploration Ph Analysis Answer Key in Everyday Life
 - Chapter 4: Student Exploration Ph Analysis Answer Key in Specific Contexts
 - Chapter 5: Conclusion
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 4. In chapter 3, this book will examine the practical applications of Student Exploration Ph Analysis Answer Key in daily life. This chapter will showcase real-world examples of how Student Exploration Ph Analysis Answer Key can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Student Exploration Ph Analysis Answer Key in specific contexts. The fourth chapter will explore how Student Exploration Ph Analysis Answer Key is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Student Exploration Ph Analysis Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.
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