

**I) Choose the correct answer**

10 Marks

- To follow up saponification of ester with alkali by conductance ( $L$ ), the following order is expected

- a)  $L_0 < L_1 < L_2$       b)  $L_0 > L_1 > L_2$       c)  $L_0 < L_2 < L_1$       d)  $L_1 > L_0 < L_2$       e) None

- In iodination of cyclohexanone, one observes that

- a) Protons are released      b)  $A$  at  $\lambda_{max}$  decreases      c) Color appears      d) anions disappear

- Normality of 83.5 wt%  $H_2SO_4$  with density =  $1.442 \text{ g/cm}^3$  ( $H=1$ ,  $O=16$ ,  $S=32$ ) is

- a) 18.1      b) 24.6      c) 10.5      d) 12.3      e) none

- A conductivity meter has a cell with two electrodes  $2 \text{ cm}^2$  each and apart  $0.1 \text{ cm}$ , the cell constant is

- a)  $2.0 \text{ cm}^{-1}$       b)  $1.0 \text{ cm}^{-1}$       c)  $10 \text{ cm}^{-1}$       d)  $20 \text{ cm}^{-1}$       e)  $0.05 \text{ cm}^{-1}$

- The  $0.1 \text{ M}$  salt solution with the highest ionic strength is

- a)  $KCl$       b)  $AgNO_3$       c)  $CaCl_2$       d)  $Al_2(SO_4)_3$       e)  $CuSO_4$

- 11.0% sucrose solution has a rotation angle of  $13.5^\circ$ . The sucrose percent at an angle of  $37.2^\circ$  is

- a) 20.9%      b) 30.3%      c) 27.6%      d) 40.2%      e) 36.3%

- For the iodination of cyclohexanone, one cannot follow the reaction kinetics by

- a) Spectrophotometer      b) Polarimeter      c) Potentiometer      d) Thiosulphate titration

- During acid hydrolysis of sucrose the rotation angle of polarimeter moves anticlockwise because

- a) Sucrose is D      b) Glucose is D      c) Glucose/Fructose is weak L      d) Both a & c      e) All a-c

- The pH of a solution composed of  $0.1 \text{ M}$  acetic acid +  $0.1 \text{ M}$  sodium acetate ( $K_a = 1.8 \times 10^{-5}$ ) is

- a) 4.0      b) 4.7      c) 5.0      d) 3.5      e) none

- The activity of  $K_2SO_4$  ( $a_{K_2SO_4}$ ) in terms of the mean activity of the salt is given by

- a)  $a_{\pm K_2SO_4}^2$       b)  $a_{\pm K_2SO_4}$       c)  $\gamma_{\pm K_2SO_4}$       d)  $a_{\pm K_2SO_4}$       e)  $a_{\pm K_2SO_4}$

II) In a saponification reaction of  $0.02 \text{ M}$  ethyl acetate with  $0.02 \text{ M}$   $KOH$ , the following conductance,  $L$ ,

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## **Table of Contents Physical Chemistry Exam Quiz Homework With Solution**

1. Understanding the eBook Physical Chemistry Exam Quiz Homework With Solution
  - The Rise of Digital Reading Physical Chemistry Exam Quiz Homework With Solution
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Chemistry Exam Quiz Homework With Solution
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Chemistry Exam Quiz Homework With Solution
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Chemistry Exam Quiz Homework With Solution
  - Personalized Recommendations

- Physical Chemistry Exam Quiz Homework With Solution User Reviews and Ratings
- Physical Chemistry Exam Quiz Homework With Solution and Bestseller Lists
- 5. Accessing Physical Chemistry Exam Quiz Homework With Solution Free and Paid eBooks
  - Physical Chemistry Exam Quiz Homework With Solution Public Domain eBooks
  - Physical Chemistry Exam Quiz Homework With Solution eBook Subscription Services
  - Physical Chemistry Exam Quiz Homework With Solution Budget-Friendly Options
- 6. Navigating Physical Chemistry Exam Quiz Homework With Solution eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Chemistry Exam Quiz Homework With Solution Compatibility with Devices
  - Physical Chemistry Exam Quiz Homework With Solution Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Chemistry Exam Quiz Homework With Solution
  - Highlighting and Note-Taking Physical Chemistry Exam Quiz Homework With Solution
  - Interactive Elements Physical Chemistry Exam Quiz Homework With Solution
- 8. Staying Engaged with Physical Chemistry Exam Quiz Homework With Solution
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Chemistry Exam Quiz Homework With Solution
- 9. Balancing eBooks and Physical Books Physical Chemistry Exam Quiz Homework With Solution
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Chemistry Exam Quiz Homework With Solution
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Chemistry Exam Quiz Homework With Solution
  - Setting Reading Goals Physical Chemistry Exam Quiz Homework With Solution
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Chemistry Exam Quiz Homework With Solution
  - Fact-Checking eBook Content of Physical Chemistry Exam Quiz Homework With Solution

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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