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# Simple Side Hustles Explained 9345 48009

**Abbey Beathan**



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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined because polyatomic ions contain multiple atoms and generally react as indivisible units any polyatomic ion that is present in both a reactant and a product in

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation must satisfy to be consistent with the law of conservation of matter

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web because of the law of conservation of mass balancing chemical equations gives a more accurate representation on what s happening in a reaction the accuracy of the ratios of moles is important in calculating the theoretical yield and how much

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide  $\text{CaO}$  and  $\text{H}_2\text{O}$  to form aqueous calcium hydroxide  $\text{Ca(OH)}_2$  is it possible to balance the equation if you incorrectly identify the product as  $\text{CaOH}$  and if so what is the equation

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web balance the following chemical equation  $\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem do 4 problems

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