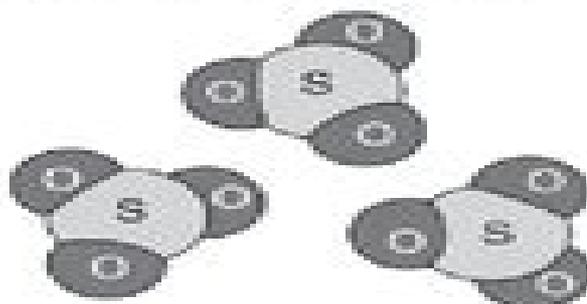


Name _____ Class _____ Date _____

1 In the space below, draw a diagram of a methane molecule, which has the formula of CH₄.

(Total for Question 1 = 2 marks)

2 (a) Give the formula for the oxide of sulfur shown in the diagram.



(1)

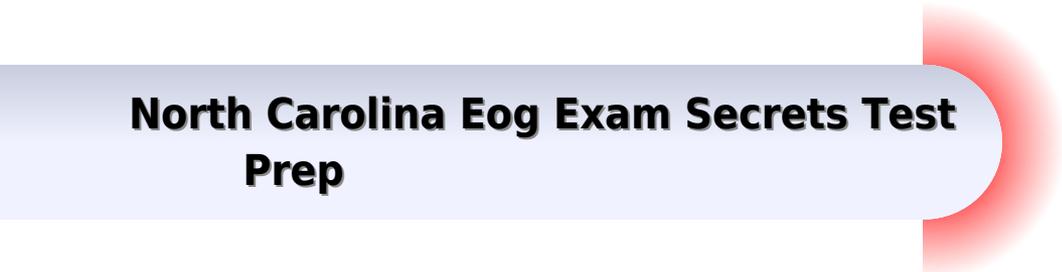
(b) 4 g of sulfur reacts completely with 6 g of oxygen to form this oxide of sulfur.
 Calculate the mass of sulfur needed to produce 30 g of the oxide.

(2)

(Total for Question 2 = 3 marks)

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