

Name: **Kyleen Mitchell**

Date: **11-5-20**

## Student Exploration: Balancing Chemical Equations

**Directions:** Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

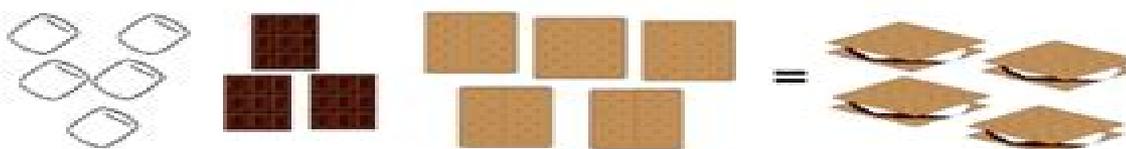
**Vocabulary:** coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

The scouts are making s'mores out of toasted marshmallows, chocolate, and graham crackers.

1. What is wrong with the image below?

What is wrong with the image below is that there are too many marshmallows and not enough chocolate and graham crackers.



2. Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, how many s'mores could you make with the ingredients shown?

Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, you could make 2 s'mores with the ingredients shown.

### Gizmo Warm-up

In a chemical reaction, **reactants** interact to form **products**. This process is summarized by a **chemical equation**. In the *Balancing Chemical Equations* Gizmo, look at the floating molecules below the initial reaction:  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ .

- How many atoms are in a hydrogen **molecule** ( $\text{H}_2$ )?
- How many atoms are in an oxygen molecule ( $\text{O}_2$ )?
- How many hydrogen and oxygen atoms are in a water molecule ( $\text{H}_2\text{O}$ )?
- In general, what does a **subscript** (such as the "2" in  $\text{H}_2$ ) tell you about the molecule?

2

2

3

In general, a subscript like "2" in  $\text{H}_2$  tells you that there are 2 H's in that molecule.



# Balancing Chemical Equations Gizmo Answers

**Xiaolong Qi**



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