

Basic Stoichiometry Post-Lab Homework Exercises

1. Load the "Reactants, Products, and Leftovers" simulation and work through each of the levels of the Game! At <http://phet.colorado.edu/> or googling "phet." You may have to download or update Java on your computer.

Show your work for all calculations. Remember: Labels, units, sig figs and cancellation!

2. For the reaction $P_4 + 6Cl_2 \rightarrow 4PCl_3$, determine how many moles of chlorine Cl_2 would be needed to react with 4.0 moles of phosphorus P_4 to entirely use up all the phosphorus. 4) _____

3. If 3.00 moles of P_4 reacted with 22.0 moles Cl_2 according to the above reaction, determine:

- a) How many moles PCl_3 are produced a) _____
- b) How many moles of P_4 are left in excess after the reaction (if any) b) _____
- c) How many moles of Cl_2 are left in excess after the reaction (if any) c) _____

In reality, reactants don't have to react in perfect whole-numbers of moles. In a two-reactant synthesis reaction, usually **one reactant gets entirely used up (and determines how much product is made)**, even if that means using fractions of a mole of reactant. For instance, when solid, metallic aluminum Al and red, liquid bromine Br_2 are brought together, they make a white solid according to the reaction $2Al + 3Br_2 \rightarrow 2AlBr_3$. If 5.0 moles of aluminum Al was reacted with 10 moles bromine Br_2 , all five moles of aluminum would react, with only 7.5 moles bromine. (2:3 mole ratio) This would produce only 5.0 moles of $AlBr_3$, leaving 2.5 moles of excess Br_2 behind.

5. Now assume 3.0 moles Al and 4.0 moles Br_2 react.

- a) How much (in moles) $AlBr_3$ gets produced? a) _____
- b) Which chemical must be the *excess reactant*? b) _____
- c) Which chemical is the *limiting reactant*? c) _____
- d) If all the limiting reactant gets used up, how much of the excess reactant is left? d) _____

Basic Stoichiometry Post Lab Homework Exercises

Answer Key

Jianjun Gao



Basic Stoichiometry Post Lab Homework Exercises Answer Key:

Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Basic Stoichiometry Post Lab Homework Exercises Answer Key**

In some sort of inundated with monitors and the cacophony of instantaneous interaction, the profound energy and emotional resonance of verbal beauty often diminish into obscurity, eclipsed by the continuous barrage of noise and distractions. However, situated within the lyrical pages of **Basic Stoichiometry Post Lab Homework Exercises Answer Key**, a interesting perform of literary beauty that pulses with fresh emotions, lies an wonderful trip waiting to be embarked upon. Composed by a virtuoso wordsmith, this exciting opus books visitors on a mental odyssey, softly exposing the latent potential and profound impact stuck within the complex web of language. Within the heart-wrenching expanse with this evocative examination, we can embark upon an introspective exploration of the book is main themes, dissect their fascinating publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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Table of Contents Basic Stoichiometry Post Lab Homework Exercises Answer Key

1. Understanding the eBook Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - The Rise of Digital Reading Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Basic Stoichiometry Post Lab Homework Exercises Answer Key

- Personalized Recommendations
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key User Reviews and Ratings
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key and Bestseller Lists
5. Accessing Basic Stoichiometry Post Lab Homework Exercises Answer Key Free and Paid eBooks
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key Public Domain eBooks
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key eBook Subscription Services
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key Budget-Friendly Options
 6. Navigating Basic Stoichiometry Post Lab Homework Exercises Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key Compatibility with Devices
 - Basic Stoichiometry Post Lab Homework Exercises Answer Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Highlighting and Note-Taking Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Interactive Elements Basic Stoichiometry Post Lab Homework Exercises Answer Key
 8. Staying Engaged with Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Basic Stoichiometry Post Lab Homework Exercises Answer Key
 9. Balancing eBooks and Physical Books Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Basic Stoichiometry Post Lab Homework Exercises Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Setting Reading Goals Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Basic Stoichiometry Post Lab Homework Exercises Answer Key

- Fact-Checking eBook Content of Basic Stoichiometry Post Lab Homework Exercises Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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