

Name _____ Date _____ Class _____

CHAPTER

15

STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

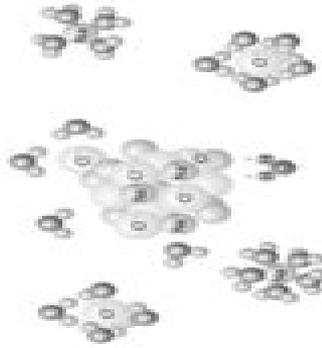
Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solute

Air is a(n) **(1)** solution of oxygen gas dissolved in nitrogen gas. The oxygen is an **(2)** solute and nitrogen is the **(3)** solvent. Because oxygen gas dissolves in a solvent, oxygen gas is an **(4)** soluble substance. A substance that does not dissolve is **(5)** insoluble. **(6)** Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** miscible. However, if one liquid is insoluble in another, the liquids are **(8)** immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



9. Hydration is hydration in which the solvent is water. What is hydration?
Solution is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER

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Section 15.1 continued

16. An sodium chloride dissolves in water. What happens to the sodium and chloride ions? The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the orientation of the water molecules around the sodium ions and chloride ions. Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

19. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

- true** 14. The overall energy change that occurs when a solution forms is called the heat of solution.
- false** 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.
- false** 16. Solvation continues as long as the solvation rate is less than the crystallization rate.
- true** 17. In a saturated solution, solvation and crystallization are in equilibrium.
- true** 18. Additional water can be dissolved in an unsaturated solution.
- true** 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Wendy Micham



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Study Guide for Use with a Child's World Wendy Micham, 2003-08 Includes chapter summaries outlines quizzes topics for thought and discussion and suggestions for further reading for each of the 17 chapters of A child's world updated 9th ed

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Thibodeau,2019-09-28 Get a solid understanding of the human body Using simple conversational language and vivid animations and illustrations Structure Function of the Body 16th Edition introduces the normal structure and function of the human body and what the body does to maintain homeostasis To help make difficult A P concepts easy to understand this new edition features thoroughly revised content and review questions which reflect the most current information available and a unique 22 page semi transparent insert of the human body Plus Connect It boxes throughout directly correlate to online content giving you additional clinical and scientific insights essential to patient care 22 page Clear View of the Human Body is a unique full color semi transparent insert depicting the human body male and female in layers Conversational and clear writing style makes content easy to read and understand Full color design contains more than 400 drawings and photos Updated study tips sections at the beginning of each chapter help break down difficult topics and guide you on how to best use book features to their advantage Questions for student review are found throughout the chapters and cover critical thinking open ended fill in the blank matching multiple choice and other question formats Special boxes such as Health and Well Being boxes Clinical Application boxes Research and Trends boxes and more help you apply what you have learned to your future career Language of Science and Medicine section in each chapter includes key terms word parts and pronunciations to place a greater focus on medical terminology Resources on the Evolve companion website include Animation Direct audio summaries audio glossary a new online coloring book review questions and FAQs NEW Thoroughly revised chapters illustrations and review questions reflect the most current information available NEW Connect It boxes refer you to online content providing additional clinical and scientific insights NEW A P contributors join Dr Patton to enhance the content and bring additional perspectives to the book [Mastery Study Guide Into Psychology](#) Tom Bond,Coon,1991-09 [Exploring Management](#) John R. Schermerhorn, Jr.,Daniel G. Bachrach,2017-10-16 Exploring Management supports teaching and learning of core management concepts by presenting material in a straightforward conversational style with a strong emphasis on application With a focus on currency high interest examples and pedagogy that encourages critical thinking and personal reflection Exploring Management is the perfect balance between what students need and what instructors want **Human Form, Human Function** Kerry L. Hull,2011 Designed to accompany Human Form Human Function this student workbook offers chapter overviews chapter objectives focus questions mastery tests study activities and mastery test answers *World History* ,2000 *Introduction to Psychology* Dennis Coon,1986 [Ate Wh](#) Holt Rinehart & Winston,2000 [Psychology](#) Stephen F. Davis,Joseph J. Palladino,2002 For Introductory Psychology courses This text offers an exploration of classic and contemporary research on current and emerging topics in the field and encourages students to recognize first hand how psychological concepts have meaning in their own daily lives **Study Guide** George Karleskint,1991 **Modules** ,2005 [Mastering Psychology and Life](#) Karl Alfred Minke,1979 [Human Biology](#) James Trefil,2005 **Modules** McDougal Littell Incorporated,2005

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