

## 10

## CHEMICAL QUANTITIES

## Chapter Test A

## A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description on the line.

## Column A

- \_\_\_\_\_ 1. standard temperature and pressure
- \_\_\_\_\_ 2. molar volume
- \_\_\_\_\_ 3. representative particle
- \_\_\_\_\_ 4. molar mass
- \_\_\_\_\_ 5. atomic mass
- \_\_\_\_\_ 6. Avogadro's number
- \_\_\_\_\_ 7. mole (mol)
- \_\_\_\_\_ 8. empirical formula
- \_\_\_\_\_ 9. percent composition
- \_\_\_\_\_ 10. molar mass

## Column B

- a. the volume occupied by a mole of any gas at STP (22.4 L)
- b. the mass in amu of an atom of an element
- c. the number of representative particles in a mole of a substance
- d. the percent by mass of each element in a compound
- e. the mass of a mole of a substance
- f. the formula that gives the lowest whole-number ratio of the elements in a compound
- g. an atom, a formula unit, or a molecule
- h. the mass in grams of a mole of any pure substance
- i.  $6.02 \times 10^{23}$  representative particles of a substance
- j.  $0^\circ\text{C}$  and 101.3 kPa

## B. Multiple Choice

Choose the best answer and write its letter on the line.

- \_\_\_\_\_ 11. How many hydrogen atoms are in 4 molecules of isopropyl alcohol,  $\text{C}_3\text{H}_7\text{O}$ ?
- a.  $4 \times (6.02 \times 10^{23})$
- b. 4
- c. 32
- d.  $32 \times (6.02 \times 10^{23})$
- \_\_\_\_\_ 12. The mass of a mole of NaCl is its:
- a. molar mass.
- b. empirical formula.
- c. gram atomic mass.
- d. atomic mass.

# Chapter 10 Assessment Chemistry Chemical Quantities Answers

**G Thomas**



**Chapter 10 Assessment Chemistry Chemical Quantities Answers:**

This book delves into Chapter 10 Assessment Chemistry Chemical Quantities Answers. Chapter 10 Assessment Chemistry Chemical Quantities Answers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chapter 10 Assessment Chemistry Chemical Quantities Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
  - Chapter 1: Introduction to Chapter 10 Assessment Chemistry Chemical Quantities Answers
  - Chapter 2: Essential Elements of Chapter 10 Assessment Chemistry Chemical Quantities Answers
  - Chapter 3: Chapter 10 Assessment Chemistry Chemical Quantities Answers in Everyday Life
  - Chapter 4: Chapter 10 Assessment Chemistry Chemical Quantities Answers in Specific Contexts
  - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Chapter 10 Assessment Chemistry Chemical Quantities Answers. The first chapter will explore what Chapter 10 Assessment Chemistry Chemical Quantities Answers is, why Chapter 10 Assessment Chemistry Chemical Quantities Answers is vital, and how to effectively learn about Chapter 10 Assessment Chemistry Chemical Quantities Answers.
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6. In chapter 5, the author will draw a conclusion about Chapter 10 Assessment Chemistry Chemical Quantities Answers. The final chapter will summarize the key points that have been discussed throughout the book.  
The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 10 Assessment Chemistry Chemical Quantities Answers.

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