

## Chapter 3, Lesson 1: What is Density?

### *Key Concepts*

- Density is a characteristic property of a substance.
- The density of a substance is the relationship between the mass of the substance and how much space it takes up (volume).
- The mass of atoms, their size, and how they are arranged determine the density of a substance.
- Density equals the mass of the substance divided by its volume;  $D = m/v$ .
- Objects with the same volume but different mass have different densities.

### *Summary*

Students will observe a copper and an aluminum cube of the same volume placed on a balance. They will see that the copper has a greater mass. Students will try to develop an explanation, on the molecular level, for how this can be. Students are then given cubes of different materials that all have the same volume. Students determine the density of each cube and identify the substance the cube is made from.

### *Objective*

Students will be able to calculate the density of different cubes and use these values to identify the substance each cube is made of. Students will be able to explain that the size, mass, and arrangement of the atoms or molecules of a substance determines its density.

### *Evaluation*

The activity sheet will serve as the "Evaluate" component of each S-E lesson plan. The activity sheets are formative assessments of student progress and understanding. A more formal summative assessment is included at the end of each chapter.

### *Safety*

Make sure you and your students wear properly fitting goggles.

### *Materials for Each Group*

- Cubes marked A–H that you will share with other groups
- Balance that can measure in grams
- Calculator

### *Materials for the Demonstration*

- Copper cube and aluminum cube of the same volume
- Balance

# Chapter 3 Lesson 1 What Is Density

**Emilie Sanchez**



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### **Table of Contents Chapter 3 Lesson 1 What Is Density**

1. Understanding the eBook Chapter 3 Lesson 1 What Is Density
  - The Rise of Digital Reading Chapter 3 Lesson 1 What Is Density
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 3 Lesson 1 What Is Density
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 3 Lesson 1 What Is Density
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 3 Lesson 1 What Is Density
  - Personalized Recommendations
  - Chapter 3 Lesson 1 What Is Density User Reviews and Ratings
  - Chapter 3 Lesson 1 What Is Density and Bestseller Lists
5. Accessing Chapter 3 Lesson 1 What Is Density Free and Paid eBooks
  - Chapter 3 Lesson 1 What Is Density Public Domain eBooks
  - Chapter 3 Lesson 1 What Is Density eBook Subscription Services

- Chapter 3 Lesson 1 What Is Density Budget-Friendly Options
- 6. Navigating Chapter 3 Lesson 1 What Is Density eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 3 Lesson 1 What Is Density Compatibility with Devices
  - Chapter 3 Lesson 1 What Is Density Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 3 Lesson 1 What Is Density
  - Highlighting and Note-Taking Chapter 3 Lesson 1 What Is Density
  - Interactive Elements Chapter 3 Lesson 1 What Is Density
- 8. Staying Engaged with Chapter 3 Lesson 1 What Is Density
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 3 Lesson 1 What Is Density
- 9. Balancing eBooks and Physical Books Chapter 3 Lesson 1 What Is Density
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 3 Lesson 1 What Is Density
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 3 Lesson 1 What Is Density
  - Setting Reading Goals Chapter 3 Lesson 1 What Is Density
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 3 Lesson 1 What Is Density
  - Fact-Checking eBook Content of Chapter 3 Lesson 1 What Is Density
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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