

28.2. Three methods for improving resolution are: (1) Adjustment of  $k_a'$  and  $k_b'$  by employing a multi-component mobile phase and varying the ratio of the components to find an optimal mixture, (2) variation in the chemical composition of the solvent system in such a way as to make  $\alpha$  larger, and (3) by employing a different packing in which  $\alpha$  is greater.

28.3. In partition chromatography  $k'$  is conveniently varied using a two or more component solvent system and varying the ratio of the solvents.

28.4. In gas-liquid chromatography  $\alpha$  is generally varied through the choice of column packing. For liquid chromatography both column packing and chemical composition of the mobile phase can be varied to yield better  $\alpha$  values.

28.5. In adsorption chromatography on an alumina packing it is generally best to increase the polarity of the mobile phase as the elution proceeds. Thus the ratio of acetone to benzene should be increased as elution proceeds.

28.13. In ion-exchange chromatography, ions are separated based on their exchange rate with the stationary phase. With size-exclusion chromatography, neutral molecules are separated based on the size of the molecule.

28.20. Because the peak widths are equal, we can assume that  $R_s = \frac{T_r(b) - T_r(a)}{W}$ , which allows us to use equation 26-26, which is

$$N = 16R_s^2 \left( \frac{\alpha}{\alpha - 1} \right)^2 \left( \frac{1 + k_B}{k_B} \right)^2$$

To find  $\alpha$ , we need to find  $k_B$  and  $k_A$ , using the following equation

$$k = \frac{T_r - T_m}{T_m}$$

and converting the retention times to either all seconds or all minutes, giving the following values:

$k(a)$	$k(b)$	$\alpha$
5	5.166667	1.033333

$R_s$	$N$
0.5	5476
0.75	12321
0.9	17742.24
1	21904
1.1	26503.84
1.25	34225
1.5	49284
1.75	67081

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