

### Concentration

Mass

Volume

### Concentration of solution

#### Quantities

- 1. Mass concentration of solution
- 2. Molarity
- 3. Molar mass

#### Mass

How much mass (m) you provide in preparing the concentration of a substance with a certain volume (V) and density (ρ) is the amount in grams of a substance in solution of certain concentration.

#### Calculations

How to calculate mass (m) and volume (V) of a solution

- 1. Calculate concentration of solution (g/cm<sup>3</sup>) of a substance (m) with a certain volume (V) of a solution
- 2. Calculate volume (V) of a solution with a certain mass (m) of a substance

#### Calculating the volume

Volume (V) can be calculated by dividing mass (m) by density (ρ) of a substance. For example, if you have 100g of a substance with a density of 2g/cm<sup>3</sup>, the volume (V) of the solution is 50cm<sup>3</sup>.

Mass (m) of a substance in a solution can be calculated by multiplying the volume (V) of the solution by the density (ρ) of the substance. For example, if you have 50cm<sup>3</sup> of a solution with a density of 2g/cm<sup>3</sup>, the mass (m) of the substance is 100g.

$$\text{Concentration (g/cm}^3\text{)} = \frac{\text{Mass (m)}}{\text{Volume (V)}}$$

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# Ssd Solution Formula

**Rosina Ehmann**



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