

Chemistry Worksheet Class 11 on Chapter 13 Hydrocarbons - Set 1

Q-1: Which of the following is the chemical formula of marsh gas?

- a) C_2H_6
- b) CH_4
- c) C_2H_2
- d) C_4H_{10}

Q-2: What would the alkane's formula be if one of the hydrogen atoms in propane was replaced by an ethyl group?

Q-3: Calculate the number of sigma and pi bonds in the following compounds:

- a) 4-Ethyl-2-methylhexane
- b) 3,3-Diethyl-5-isopropyl-4-methyloctane
- c) 4-Ethyl-2,5,7-decatriene

Q-4: Give the IUPAC name for the following:

- a) $CH_3-CH=CH_2$
- b) $CH_2=C(CH_3)_2$
- c) $CH_3-CH_2-C(CH_3)=CH_2$
- d) $CH_3=CH-CH=CH_2$
- e) $CH_3-CH_2-CH(CH_2-CH_3)-CH_2-CH(CH_3)-CH_2-CH_3$

Q-5: What happens to the electron cloud of carbon-hydrogen bonds in an ethane molecule if the configuration is changed from staggered to eclipsed?

Q-6: Which of the following conformations has hydrogen atoms attached to two carbon atoms as close together as possible?

- a) Staggered
- b) Eclipsed
- c) Skew
- d) Gauche

Q-7: What type of isomerism do the compounds listed below exhibit?

Chapter 13 Chemistry Test Answers

AN Whitehead



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Table of Contents Chapter 13 Chemistry Test Answers

1. Understanding the eBook Chapter 13 Chemistry Test Answers
 - The Rise of Digital Reading Chapter 13 Chemistry Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 13 Chemistry Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 13 Chemistry Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 13 Chemistry Test Answers
 - Personalized Recommendations

- Chapter 13 Chemistry Test Answers User Reviews and Ratings
- Chapter 13 Chemistry Test Answers and Bestseller Lists
- 5. Accessing Chapter 13 Chemistry Test Answers Free and Paid eBooks
 - Chapter 13 Chemistry Test Answers Public Domain eBooks
 - Chapter 13 Chemistry Test Answers eBook Subscription Services
 - Chapter 13 Chemistry Test Answers Budget-Friendly Options
- 6. Navigating Chapter 13 Chemistry Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 13 Chemistry Test Answers Compatibility with Devices
 - Chapter 13 Chemistry Test Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 13 Chemistry Test Answers
 - Highlighting and Note-Taking Chapter 13 Chemistry Test Answers
 - Interactive Elements Chapter 13 Chemistry Test Answers
- 8. Staying Engaged with Chapter 13 Chemistry Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 13 Chemistry Test Answers
- 9. Balancing eBooks and Physical Books Chapter 13 Chemistry Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 13 Chemistry Test Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 13 Chemistry Test Answers
 - Setting Reading Goals Chapter 13 Chemistry Test Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 13 Chemistry Test Answers
 - Fact-Checking eBook Content of Chapter 13 Chemistry Test Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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