

6. What is meant by atomicity? Name a diatomic element.

Solution:

A set of atoms of the same type together forms a molecule of the element. The number of atoms in a molecule of an element is called its atomicity.

Examples of diatomic elements are H_2 – Hydrogen, O_2 – Oxygen, N_2 – Nitrogen

7. a) Explain the terms valency and variable valency

b) How are the elements with variable valency named? Explain with an example.

Solution:

a) Valency is the capacity of an atom to lose, gain or share atoms during a chemical reaction is called its valency. Sometimes atom of an element can lose more electron than they are present which means they lose electron from its penultimate shell. Such an element is said to exhibit variable valency.

b) If an element exhibits two different positive valencies, then

1. For the lower valency, use the suffix -OUS at the end of the name of the metal
2. For the higher valency, use the suffix -IC at the end of the name of the metal.

8. Give the formula and valency of

- a) aluminate
- b) chromate
- c) aluminium
- d) cupric

Solution:

Name	Formula	Valency
Aluminate	AlO_2	-2
Chromate	CrO_4	-2
Aluminium	Al	+3
Cupric	Cu	+2

9. a) What is a chemical formula?

b) What is the significance of a formula? Give an illustrate.

Solution:

a) Chemical formula is a symbolic representation of the number of atoms present in a molecule of that substance.

b) Significance of Chemical Formula

Chemical formula is very important in finding information about chemical compounds as it tells us about the elements and the number of atoms in a substance

Example - Salt - $NaCl$, ethanol C_2H_6O because the molecule of ethanol contains two Carbon, 6 Hydrogen and 1 Oxygen atom.

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