

Chemical Equations and Reactions

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

- | | |
|----------------------------------|-----------------------------------|
| <u> c </u> synthesis | (a) $AX + BY \rightarrow AY + BX$ |
| <u> d </u> decomposition | (b) $A + BX \rightarrow AX + B$ |
| <u> b </u> single-displacement | (c) $A + B \rightarrow AX$ |
| <u> a </u> double-displacement | (d) $AX \rightarrow A + X$ |

2. c In the reaction described by the equation $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$, iron has been replaced by
 (a) nitrate. (c) **aluminum.**
 (b) water. (d) nitrogen.

3. a Of the following chemical equations, the only reaction that is both synthesis and combustion is
 (a) **$C(s) + O_2(g) \rightarrow CO_2(g)$.**
 (b) $2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.
 (c) $6CO_2(g) + 6H_2O(g) \rightarrow C_6H_{12}O_6(aq) + 6O_2(g)$.
 (d) $C_6H_{12}O_6(aq) + 6O_2(g) \rightarrow 6CO_2(aq) + 6H_2O(l)$.

4. b Of the following chemical equations, the only reaction that is both combustion and decomposition is
 (a) $S(s) + O_2(g) \rightarrow SO_2(g)$.
 (b) **$2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.**
 (c) $2H_2O_2(l) \rightarrow 2H_2O(l) + O_2(g)$.
 (d)

5. Identify the products when the following substances decompose:

- Its separate elements** _____ a binary compound
- Metal oxide + water** _____ most metal hydroxides
- Metal oxide + carbon dioxide** _____ a metal carbonate
- Water + sulfur dioxide** _____ the acid H_2SO_3

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

Chemical Equations Reactions Section 2 Answers

Mark Steyvers



Chemical Equations Reactions Section 2 Answers:

The book delves into Chemical Equations Reactions Section 2 Answers. Chemical Equations Reactions Section 2 Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemical Equations Reactions Section 2 Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemical Equations Reactions Section 2 Answers
- Chapter 2: Essential Elements of Chemical Equations Reactions Section 2 Answers
- Chapter 3: Chemical Equations Reactions Section 2 Answers in Everyday Life
- Chapter 4: Chemical Equations Reactions Section 2 Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chemical Equations Reactions Section 2 Answers. This chapter will explore what Chemical Equations Reactions Section 2 Answers is, why Chemical Equations Reactions Section 2 Answers is vital, and how to effectively learn about Chemical Equations Reactions Section 2 Answers.

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6. In chapter 5, the author will draw a conclusion about Chemical Equations Reactions Section 2 Answers. This chapter will summarize the key points that have been discussed throughout the book.

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