

# "CHEMICAL BONDING"



8 electron



6 electron

↓  
Short octet  
Group IIIA element  
in compound form  
show short octet



10 electron  
↓  
extended octet

- a) NH<sub>3</sub>
- b) SO<sub>2</sub>
- c) AlCl<sub>3</sub>
- d) B and C

- SHORT OCTETS ARE ALWAYS LEWIS ACID.
- EXTENDED OCTETS MAY OR MAY NOT BE LEWIS BASE.

## ENERGETICS OF BOND FORMATION:

### Formation

$\Delta H = -ve$

Energy released

So overallly energy less

So stability occurs.

Bond length

or Bond distance

or compromise bond distance.

### Breakage

$\Delta H = +ve$

Heat absorbed.

Overall energy increases

stability loses.

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