



The Chemistry of Life

Chapter Test A

Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

- _____ 1. Of the four macromolecules (carbohydrates, lipids, proteins, and nucleic acids), the most readily broken down for energy is
- carbohydrates
 - lipids
 - proteins
 - nucleic acids
- _____ 2. Of the four macromolecules (carbohydrates, lipids, proteins, and nucleic acids), the most readily broken down for energy is
- carbohydrates
 - lipids
 - proteins
 - nucleic acids
- _____ 3. Which of the following is **not** a characteristic of water?
- low density of hydrogen-bonded water molecules
 - very small surface tension and surface area
 - universal solvent for many organic molecules
 - high density of hydrogen-bonded water molecules
- _____ 4. A molecule that is neutral in the standard state is
- uncharged
 - uncharged
 - charged
 - charged
- _____ 5. Which molecule is **not** polar, with the
- oxygen with being slightly positive and the hydrogen with being slightly negative
 - oxygen with being slightly positive
 - oxygen with being slightly negative
 - oxygen with being slightly positive and the hydrogen with being slightly positive
- _____ 6. If you are left with nothing, what are you left with?
- nothing
 - nothing
 - nothing
 - nothing
- _____ 7. A substance with a pH of 7 is called
- acidic
 - basic
 - neutral
 - amphoteric
- _____ 8. Nucleic acids are made of
- proteins
 - lipids
 - carbohydrates
 - nucleic acids
- _____ 9. Why is caffeine considered compared to other substances?
- Caffeine is a stimulant and has a high pH and is a weak acid.
 - Caffeine is a stimulant and has a high pH and is a weak base.
 - Caffeine is a stimulant and has a high pH and is a weak acid.
 - Caffeine is a stimulant and has a high pH and is a weak base.

Chapter Test Chemistry Of Life Answer Key

WJ Hussar



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