

CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfide, the roman numeral tells us (2)
 - How many atoms of Fe are in one formula unit.
 - How many sulfide ions can be obtained by the ion atom.
 - The charge on each Fe ion.
 - The total positive charge of the formula unit.
- The result of changing a subscript in a correctly written chemical formula is (1)
 - Change the number of ions represented in the formula.
 - Change the charges on the other ions in the compound.
 - Change the formula so that it no longer represents the compound.
 - Have no effect on the formula.
- Derive a generalization for when an acid name will end in the suffix -ic or -ous. If ending in -ic, the anion ended in -ate. If ending in -ous, the anion ended in -ite.
- Derive a generalization for when an acid name will begin with the prefix hydro-. If prefix is hydro-, the anion ended in -ite. This is almost always a monatomic ion.
- Fill in the blanks in the table below.

Compound Name	Formula
Acetic acid	$\text{CH}_3\text{CO}_2\text{H}$
Aluminum sulfate	$\text{Al}_2(\text{SO}_4)_3$
Lead II carbonate	PbCO_3
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Sulfuric acid	H_2SO_4
Nitrogen pentoxide	N_2O_5
Copper II carbonate	CuCO_3
Sodium sulfate	Na_2SO_4
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Mercury sulfate	HgSO_4
Nitrous acid	HNO_2
Magnesium perchlorate	$\text{Mg}(\text{ClO}_4)_2$
Iron II nitrate	$\text{Fe}(\text{NO}_3)_2$
Iron III nitrate	$\text{Fe}(\text{NO}_3)_3$
Carbon Dioxide	CO_2
Sulfur dioxide	SO_2
Sulfur pentachloride	SCl_5
Tetraphosphorus hexasulfide	P_4S_6
Carbonic acid	H_2CO_3

- Name the following compound and identify its molecular ions or acid
 - $\text{Ca}(\text{NO}_3)_2$, calcium (II) nitrate - II
 - $\text{Fe}(\text{NO}_3)_3$, Iron (III) nitrate - III
 - $\text{Fe}(\text{OH})_2$, Iron (II) hydroxide - II
 - N_2O_5 , Nitrogen pentoxide - III

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Puja Mehta



Chapter 9 Chemical Names Formulas Answers Page 221:

Reviewing **Chapter 9 Chemical Names Formulas Answers Page 221**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Chapter 9 Chemical Names Formulas Answers Page 221**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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