

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. c In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
- how many atoms of Fe are in one formula unit
 - how many sulfate ions can be attached to the iron atom
 - the charge on each Fe ion
 - the total positive charge of the formula unit
2. c The result of changing a subscript in a correctly written chemical formula is to _____
- change the number of moles represented by the formula
 - change the charges on the other ions in the compound
 - change the formula so that it no longer represents the compound it previously represented
 - have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
- | | |
|--|---|
| <u> 4 elements </u> | a. How many elements make up this compound? |
| <u> 6 oxygen atoms </u> | b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u> 21 atoms </u> | c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u> 4.2×10^{24} atoms </u> | d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$? |
4. How many atoms are present in each of these formula units?
- | | |
|---------------------|-------------------------|
| <u> 11 atoms </u> | a. $Ca(HCO_3)_2$ |
| <u> 45 atoms </u> | b. $C_{12}H_{22}O_{11}$ |
| <u> 10 atoms </u> | c. $Fe(ClO_2)_3$ |
| <u> 9 atoms </u> | d. $Fe(ClO_3)_2$ |
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
- carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
- H_2SO_3 c. What is the formula for sulfurous acid?
- phosphoric acid d. What is the name for the acid H_3PO_4 ?

Chapter 7 Chemical Formulas And Compounds Test

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[The Systematic Identification of Organic Compounds](#) Christine K. F. Hermann, Terence C. Morrill, Ralph L. Shriner, Reynold C. Fuson, 2023-03-08 The Systematic Identification of Organic Compounds A comprehensive introduction to the identification of unknown organic compounds Identifying unknown compounds is one of the most important parts of the study of chemistry From basic characteristics such as melting and or boiling point to more complex data generated through cutting edge techniques the range of possible methods for identifying unknown organic compounds is substantial The utility of a research reference which compiles known techniques and characteristics of possible compounds is clear The Systematic Identification of Organic Compounds provides such a reference designed to teach a hands on approach in the chemistry lab It takes readers step by step through the process of identifying an unknown compound and elucidating its structure from infrared nuclear magnetic resonance and mass spectra in addition to solubility characteristics

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exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed Peter P. Berlow, 1978

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Candido, Edward S. James, Ronald E. Phillips, 1987 **A Method for the identification of pure organic compounds by a systematic analytical procedure based on physical properties and chemical reactions ... v. 1, 1911** Samuel Parsons Mulliken, 1904

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The Systematic Identification of Organic Compounds Ralph L. Shriner, 1998 Dedicated to qualitative organic chemistry this book explains how to identify organic compounds through step by step instructions Topics include elemental analysis solubility infrared nuclear magnetic resonance and mass spectra classification tests and preparation of a derivative Most directions for experiments are described

in micro or mini scales Discusses chromatography distillations and the separation of mixtures Questions and problems
emphasize the skills required in identifying unknown samples Philadelphia Monthly Medical Journal ,1899 **Biennial**
Report Oregon. Board of Horticulture,1917

Chapter 7 Chemical Formulas And Compounds Test Book Review: Unveiling the Power of Words

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web preces the cantor and choir sing the introductory verse lord open thou our lips and our mouth shall shew forth thy praise god make speed to save us lord make haste to help us glory be to the father and to the son and to the holy ghost as it was in the beginning is now and ever shall be world without end amen praise ye the lord

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