

# THERMODYNAMICS

## ★ THERMODYNAMICS :

It is the combination of two words:

- Therm: Heat energy.
- Dynamics: Mechanical energy.

## ★ SCALES OF TEMPERATURE :

Temperature can be measured by the following '3' Scales.

	<u>Quantity</u>	<u>°C</u>	<u>°F</u>	<u>K</u>
1.	Boiling Point	100°C	212°F	373 K
2.	Melting Point	0°C	32°F	273 K
3.	Divisions b/w B.P & M.P	100	180	100

## ★ TEMPERATURE :

↳ Intensity or degree of hotness and coldness.

- Denoted: T.
- Nature: Scalar.
- SI unit: K.
- Dimension: [K], [Θ].
- Other units: °C, °F.
- Measurement: Thermometer.

## ★ RELATIONSHIP b/w SCALES :

$$\Rightarrow \frac{\text{Scale Reading} - \text{Freezing Point (H}_2\text{O)}}{\text{no of divisions}} = \frac{\text{Scale} \cdot R - \text{Freezing P (H}_2\text{O)}}{\text{no of divisions}}$$

$$\Rightarrow \frac{^{\circ}\text{C} - 0^{\circ}}{100} = \frac{\text{K} - 273}{100} \quad \text{and} \quad \Rightarrow$$

$$\frac{^{\circ}\text{C} - 0^{\circ}}{100} = \frac{^{\circ}\text{F} - 32^{\circ}}{180}$$

$$\frac{^{\circ}\text{C}}{10} = \frac{^{\circ}\text{F} - 32^{\circ}}{18} \Rightarrow ^{\circ}\text{C} = \frac{5}{18} (^{\circ}\text{F} - 32^{\circ})$$

$$\boxed{^{\circ}\text{C} = \text{K} - 273}$$

$$\boxed{^{\circ}\text{C} = \frac{5}{9} (^{\circ}\text{F} - 32)} \quad \text{or}$$

$$\boxed{^{\circ}\text{F} = \frac{9}{5} (^{\circ}\text{C} + 32)}$$

(1)

→ At -40, °C and °F readings are equal.

→ At 574.6, °F and K readings are equal.

# Engineering Thermodynamics Notes

**D Keegan**



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in Heat Transfer*, 1982-06-24 *Advances in Heat Transfer* *Lecture Notes for Thermodynamics for Engineers and Chemists*  
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J. Assael, Geoffrey C. Maitland, Thomas Maskow, Urs von Stockar, William A. Wakeham, Stefan Will, 2022-08-05 CRC Press is  
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mechanical theory measurement of thermodynamic quantities and their relationships phase behavior in single and  
multicomponent systems electrochemistry and chemical and biochemical reaction equilibria The later chapters explore  
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### **Table of Contents Engineering Thermodynamics Notes**

1. Understanding the eBook Engineering Thermodynamics Notes
  - The Rise of Digital Reading Engineering Thermodynamics Notes
  - Advantages of eBooks Over Traditional Books
2. Identifying Engineering Thermodynamics Notes
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Engineering Thermodynamics Notes
  - User-Friendly Interface
4. Exploring eBook Recommendations from Engineering Thermodynamics Notes

- Personalized Recommendations
  - Engineering Thermodynamics Notes User Reviews and Ratings
  - Engineering Thermodynamics Notes and Bestseller Lists
5. Accessing Engineering Thermodynamics Notes Free and Paid eBooks
    - Engineering Thermodynamics Notes Public Domain eBooks
    - Engineering Thermodynamics Notes eBook Subscription Services
    - Engineering Thermodynamics Notes Budget-Friendly Options
  6. Navigating Engineering Thermodynamics Notes eBook Formats
    - ePub, PDF, MOBI, and More
    - Engineering Thermodynamics Notes Compatibility with Devices
    - Engineering Thermodynamics Notes Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Engineering Thermodynamics Notes
    - Highlighting and Note-Taking Engineering Thermodynamics Notes
    - Interactive Elements Engineering Thermodynamics Notes
  8. Staying Engaged with Engineering Thermodynamics Notes
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Engineering Thermodynamics Notes
  9. Balancing eBooks and Physical Books Engineering Thermodynamics Notes
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Engineering Thermodynamics Notes
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Engineering Thermodynamics Notes
    - Setting Reading Goals Engineering Thermodynamics Notes
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Engineering Thermodynamics Notes

- Fact-Checking eBook Content of Engineering Thermodynamics Notes
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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