

Heat and Material Balance

| Stream Description | | Flue Gas to DCC | | Neutralized Excess DCC Water | | Demin Water (Note 2) | | 50 wt% NaOH (Note 2) | | Scrubbing Solution Blowdown from DCC | |
|---|-------|-----------------|---------|------------------------------|--------|----------------------|--------|----------------------|-------|--------------------------------------|-------|
| Stream Numbers | | 101 | | 107 | | 109 | | 110 | | 115 | |
| Temperature, °C | | 48.7 | | 48.6 | | 15.0 | | 20.0 | | 27.9 | |
| Pressure, bar(a) | | 1.00 | | 4.00 | | 1.35 | | 1.35 | | 4.00 | |
| Component Flow | MM | kgmo/hr | mol% | kgmo/hr | mol% | kgmo/hr | mol% | kgmo/hr | mol% | kgmo/hr | mol% |
| H ₂ O | 18.02 | 3,491 | 11.2% | 2,436 | 100.0% | 8.0 | 100.0% | 2.75 | 69.0% | 18 | 96.7% |
| CO ₂ | 44.01 | 4,268 | 13.7% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| N ₂ | 28.02 | 22,086 | 70.9% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| Ar | 39.95 | 249 | 0.8% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| O ₂ | 32.00 | 1,059 | 3.4% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| SO ₂ | 64.06 | 0.07 | 2 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| SO ₃ | 64.06 | 0.44 | 14 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| NO ₂ | 46.01 | 0.05 | 2 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| NO | 30.01 | 1.46 | 47 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| NH ₃ | 17.03 | 0.210 | 7 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| HCl | 36.46 | 0.06 | 2 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| HF | 20.01 | 0.02 | 1 ppmv | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| Sodium Salts | - | 0 | 0.0% | 0.11 | 0.0% | 0 | 0.0% | 1.25 | 31.0% | 0.59 | 3.0% |
| Ammonium Salts | - | 0 | 0.0% | 0.20 | 0.0% | 0 | 0.0% | 0 | 0.0% | 0 | 0.0% |
| Total Molar Flow, kgmo/hr | | 31,155 | | 2,436 | | 8.0 | | 4.0 | | 19 | |
| Total Mass Flow, kg/hr | | 913,340 | | 43,910 | | 140 | | 100 | | 410 | |
| Molecular Weight | | 29.3 | | 18.0 | | 18.0 | | 23.8 | | 21.0 | |
| Density, kg/m ³ | | 1,099 | | 999.8 | | 999.3 | | 1,528.1 | | 1,150.1 | |
| Liquid Flow, m ³ /hr | | | | 44.3 | | 0.14 | | 0.06 | | 0.34 | |
| Vapor Flow, m ³ _{dry} /hr (dry) | | 620,061 | | | | | | | | | |
| Vapor Flow, m ³ _{wet} /hr (wet) | | 696,308 | | | | | | | | | |
| Vapor Flow, m ³ /hr (actual) | | 631,095 | | | | | | | | | |
| pH | | | | 7.0 | | 7.2 | | 13.7 | | 8.2 | |

Energy Balance Chemical Engineering Spreadsheet

Max S. Peters, Klaus D. Timmerhaus



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Chemical Engineering Design Ray Sinnott, 2005-07-01 Chemical Engineering Design is one of the best known and widely adopted texts available for students of chemical engineering. It deals with the application of chemical engineering principles to the design of chemical processes and equipment. Revised throughout the fourth edition covers the latest aspects of process design, operations, safety, loss prevention, and equipment selection among others. Comprehensive and detailed, the book is supported by problems and selected solutions. In addition, the book is widely used by professionals as a day-to-day reference. Best-selling chemical engineering text. Revised to keep pace with the latest chemical industry changes designed to see students through from undergraduate study to professional practice. End of chapter exercises and solutions. *Material And Energy Balances For Engineers And Environmentalists* Colin William Oloman, 2009-05-21 Material and energy balances are fundamental to many engineering disciplines and have a major role in decisions related to sustainable development. This text, which covers the substance of corresponding undergraduate courses, presents the balance concepts and calculations in a format accessible to students, engineering professionals, and others who are concerned with the material and energy future of our society. Following a review of the basic science and economics, the text focuses on material and energy accounting in batch and continuous operations with emphasis on generic process units, flow sheets, stream tables, and spreadsheet calculations. There is a unified approach to reactive and non-reactive energy balance calculations, plus chapters dedicated to the general balance equation and simultaneous material and energy balances. Seventy worked examples show the elements of process balances and connect them with the material and energy concerns of the 21st century. **Chemical Engineering Design** Gavin Towler, Ray Sinnott, 2007-11-26 Bottom line: For a holistic view of chemical engineering design, this book provides as much if not more than any other book available on the topic. Extract from Chemical Engineering Resources review: Chemical Engineering Design is one of the best known and widely adopted texts available for students of chemical engineering. It deals with the application of chemical engineering principles to the design of chemical processes and equipment. Revised throughout, this US edition has been specifically developed for the US market. It covers the latest aspects of process design, operations, safety, loss prevention, and equipment selection among others. Comprehensive in coverage, exhaustive in detail, it is supported by extensive problems and a separate solutions manual for adopting tutors and lecturers. In addition, the book is widely used by professions as a day-to-day reference. Provides students with a text of unmatched relevance for the Senior Design Course and Introductory Chemical Engineering Courses. Teaches commercial engineering tools for simulation and costing. Comprehensive coverage of unit operations, design, and economics. Strong emphasis on HSE issues, codes, and standards including API, ASME, and ISA design codes and ANSI standards. 108 realistic commercial design projects from diverse industries. **Encyclopedia of Chemical Processing and Design** John J. McKetta Jr, 1992-11-25 Written by engineers for engineers, with over 150 International Editorial Advisory Board members, this highly lauded resource

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