



# COVALENT BONDING

## Practice Problems

Do your answers to the following problems.

### SECTION 8.1 MOLECULAR COMPOUNDS

- Classify each of the following as either ionic or molecular.
  - $\text{NaCl}$
  - $\text{H}_2\text{O}$
  - $\text{HCl}$
  - $\text{KNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}_2$
  - $\text{H}_2$
  - $\text{O}_2$

- Classify each of the following as either ionic or molecular compound. Write the name of the compound.
- $\text{K}_2\text{SO}_4$
- $\text{H}_2\text{O}$
- $\text{H}_2\text{SO}_4$
- $\text{H}_2\text{O}_2$
- $\text{H}_2$
- $\text{O}_2$

### SECTION 8.2 THE NATURE OF COVALENT BONDING

- Show covalent bonding in hydrogen fluoride.  $\text{H}-\text{F}$
- Show covalent bonding in methane.  $\text{CH}_4$
- Show covalent bonding in carbon dioxide.  $\text{CO}_2$
- Show covalent bonding in acetylene.  $\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw Lewis structures for the following molecules.
  - $\text{H}_2\text{O}$
  - $\text{H}_2\text{O}_2$
  - $\text{H}_2\text{S}$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{SO}_3$
  - $\text{H}_2\text{SO}_2$
  - $\text{H}_2\text{SO}$
  - $\text{H}_2\text{S}_2\text{O}_7$
  - $\text{H}_2\text{S}_2\text{O}_8$
  - $\text{H}_2\text{S}_2\text{O}_3$
  - $\text{H}_2\text{S}_2\text{O}_4$
  - $\text{H}_2\text{S}_2\text{O}_5$
  - $\text{H}_2\text{S}_2\text{O}_6$
  - $\text{H}_2\text{S}_2\text{O}_8$
  - $\text{H}_2\text{S}_2\text{O}_9$
  - $\text{H}_2\text{S}_2\text{O}_{10}$
  - $\text{H}_2\text{S}_2\text{O}_{11}$
  - $\text{H}_2\text{S}_2\text{O}_{12}$
  - $\text{H}_2\text{S}_2\text{O}_{13}$
  - $\text{H}_2\text{S}_2\text{O}_{14}$
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  - $\text{H}_2\text{S}_2\text{O}_{99}$
  - $\text{H}_2\text{S}_2\text{O}_{100}$

### SECTION 8.3 BONDING: THE ORBITALS

- Which orbital and sub-orbital are represented by  $\text{sp}^3$ ? The diagram shows the  $\text{sp}^3$  hybrid orbitals.
- Which orbital and sub-orbital are represented by  $\text{sp}^2$ ? Draw the  $\text{sp}^2$  hybrid orbitals.

# Chapter 8 Covalent Bonding Practice Problems Answers

**Becker, Charles F. Batten**



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