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Navigation icons: Home, Back, Forward, Refresh, Print, etc.

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	A	B	C
	Frequent Shopper	Online Shopping Percentage	Next Action
	True		
1	true	0 <= < 0.15	Email coupon for 10% off
2	true	0.15 <= < 0.5	Email a link to the holiday catalog
3	true	0.5 <= < 0.75	Offer a free web page to share ideas
4	true	>= 0.75	Setup a premium account

- Expanded View
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# Blaze Advisor Documentation

**JG Myers**



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William George Bruce, William Conrad Bruce, 1920

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web oct 10 2023 mole standard unit  $6.02214076 \times 10^{23}$  in chemistry for measuring large quantities of very small entities such as atoms molecules or other specified particles the number of units in a mole also bears the name avogadro s number or avogadro s constant in honor of the italian physicist amedeo avogadro

#### *mole unit wikipedia* - Aug 08 2022

web the mole was made the seventh si base unit in 1971 by the 14th cgm 2019 redefinition of si base units before the 2019 redefinition of the si base units the mole was defined as the amount of substance of a system that contains as many elementary entities as there are atoms in 12 grams of carbon 12 the most common isotope of carbon