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 $2\text{ }Ca(OH)_2 + 1\text{ }Al_2SO_4 \rightarrow 3\text{ }CaSO_4 + 2\text{ }Al(OH)_3$ type of reaction double displacement
 $3\text{ }Mg + 1\text{ }Fe_2O_3 \rightarrow 2\text{ }Fe + 3\text{ }MgO$ type of reaction single displacement
 $4\text{ }C_2H_4 + 3\text{ }O_2 \rightarrow 2\text{ }CO_2$

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 $1\text{ }CaF_2 + 2\text{ }Na \rightarrow 2\text{ }NaF + 1\text{ }Ca$ type of chemical reaction single displacement
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