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THE
COURAGE
TO
WRITE

{ HOW WRITERS
TRANSCEND FEAR }

RALPH KEYES

author of THE WRITER'S BOOK OF HOPE

The Courage To Write How Writers Transcend Fear
Ralph Keyes

L Cohen



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thank you for the world you are one person but for our children you are the world thank you teacher

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266 section 01 introduction in this lab we observed the relationship between pressure and temperature

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between
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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of
declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and
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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in
the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the
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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates
the volume and the number of particles in each flask indicates relative amounts