

$$\text{increase} = 14 - 8 = 6 \text{ g min}^{-1}.$$

$$\text{percentage} = \frac{6}{8} \times 100\% = 75\%$$

5. We perform an experiment to measure Planck's constant. We obtain $6.30 \times 10^{-34} \text{ J s}$. The accepted value is $6.626 \times 10^{-34} \text{ J s}$. What is the percentage error in our experiment?

$$\% \text{ error} = \frac{\text{difference}}{\text{true value}} = \frac{6.30 \times 10^{-34} - 6.626 \times 10^{-34}}{6.626 \times 10^{-34}} \times 100\% = -4.9\%$$

Percentage Yield

$$\text{percentage yield} = \frac{\text{actual yield}}{\text{maximum yield}} \times 100 \%$$

1.6 g of methanol is mixed with excess ethanoic acid. After extraction the yield is 2.5 g of ester. What is the percentage yield?



$$\text{Mol Wt CH}_3\text{OH} \quad \quad \quad = 32$$

$$\text{Mol Wt CH}_3\text{COOCH}_3 \quad \quad = 74$$

Stoichiometrically

$$32 \text{ g of CH}_3\text{OH yields } 74 \text{ g of CH}_3\text{COOCH}_3$$

$$1 \text{ g of CH}_3\text{OH yields } \frac{74}{32} \text{ g of CH}_3\text{COOCH}_3$$

$$1.6 \text{ g of CH}_3\text{OH yields } \frac{74}{32} \times 1.6 \text{ g} = 3.7 \text{ g of CH}_3\text{COOCH}_3$$

$$\text{percentage yield} = \frac{2.5}{3.7} \times 100 \% = 67.6 \%$$

Skill Practice 34 Percent Yield Answers

M Carnoy



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