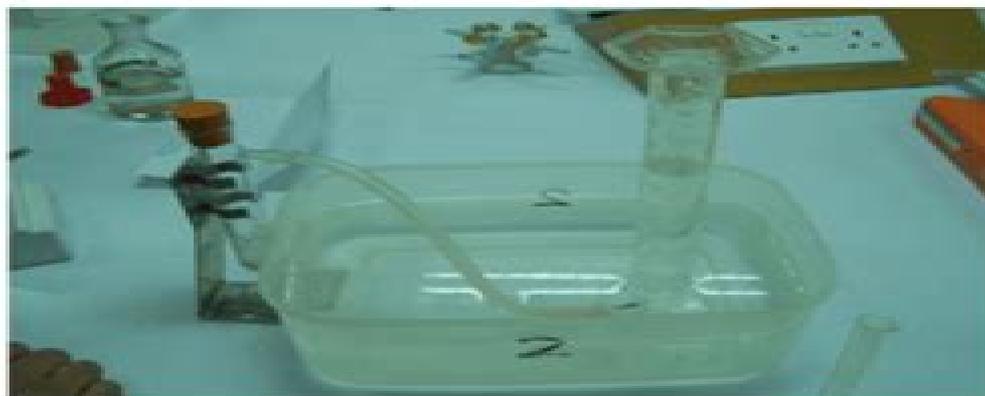
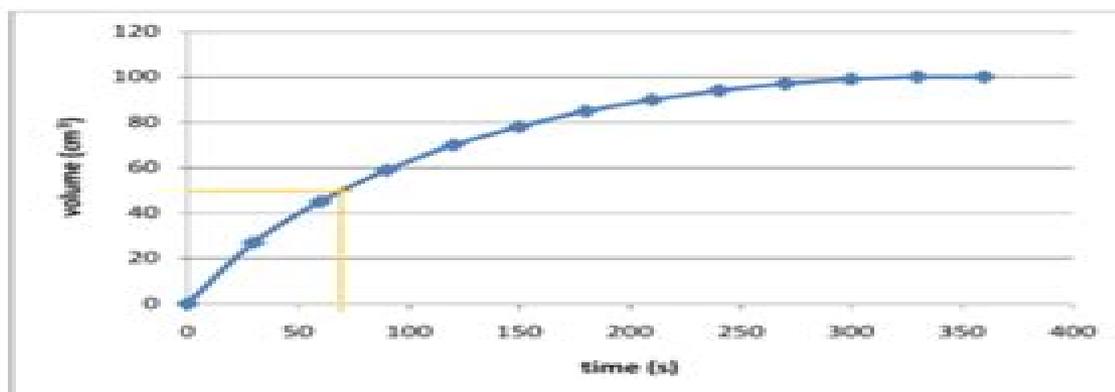


1 a) Refer to Fig 7.12 p 58 for a basic apparatus. The reaction could also be carried out in a conical flask with a side arm and the collecting vessel should be an inverted measuring cylinder of appropriate capacity to allow the volume of gas evolved to be noted at each time interval.



b)



c) The gas is being evolved most quickly at the start of the reaction (during the first 50 seconds). The concentration of reactants is greatest at this time, so the reaction proceeds most vigorously because there are more reacting particles present.

d) About 65 seconds (orange lines)

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