

4.1 EXTERNAL ASSESSMENT 2020: CHEMISTRY PAPER 1

Time allowed

- Feralal time — 10 minutes
- Working time — 95 minutes

General instructions

- Answer all questions in this question and response book.
- QCAA-approved calculator permitted.
- QCAA formula and data book provided.
- Planning paper will not be marked.

Section 1 (20 marks)

- 20 multiple choice questions

Section 2 (40 marks)

- 7 short response questions

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Answers on pages 114–127

Section 1 (20 marks)

MULTIPLE CHOICE QUESTIONS

QUESTION 1

A partly filled water bottle is sealed and left on a bench in a room with a constant temperature. After several minutes, it is noted that the water level in the bottle remains constant. In the water bottle, the rate of evaporation is

- (A) less than the rate of condensation.
- (B) greater than the rate of condensation.
- (C) equal to the rate of condensation and equal to zero.
- (D) equal to the rate of condensation but not equal to zero.

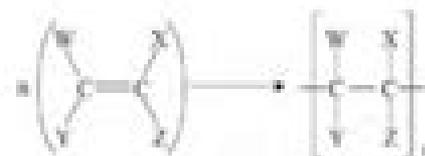
QUESTION 2



Determine which of the following statements is true for the chemical reaction.

- (A) Silver ions are reduced and act as a reducing agent.
- (B) Silver ions are oxidised and act as a reducing agent.
- (C) Magnesium atoms are reduced and act as a reducing agent.
- (D) Magnesium atoms are oxidised and act as a reducing agent.

QUESTION 3



This general chemical equation represents the following type of reaction.

- (A) addition
- (B) hydrolysis
- (C) esterification
- (D) condensation



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Table of Contents 123 Section Assessment Answers Chemistry

1. Understanding the eBook 123 Section Assessment Answers Chemistry
 - The Rise of Digital Reading 123 Section Assessment Answers Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying 123 Section Assessment Answers Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 123 Section Assessment Answers Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from 123 Section Assessment Answers Chemistry
 - Personalized Recommendations

- 123 Section Assessment Answers Chemistry User Reviews and Ratings
- 123 Section Assessment Answers Chemistry and Bestseller Lists
- 5. Accessing 123 Section Assessment Answers Chemistry Free and Paid eBooks
 - 123 Section Assessment Answers Chemistry Public Domain eBooks
 - 123 Section Assessment Answers Chemistry eBook Subscription Services
 - 123 Section Assessment Answers Chemistry Budget-Friendly Options
- 6. Navigating 123 Section Assessment Answers Chemistry eBook Formats
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 - 123 Section Assessment Answers Chemistry Enhanced eBook Features
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 - Highlighting and Note-Taking 123 Section Assessment Answers Chemistry
 - Interactive Elements 123 Section Assessment Answers Chemistry
- 8. Staying Engaged with 123 Section Assessment Answers Chemistry
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 123 Section Assessment Answers Chemistry
- 9. Balancing eBooks and Physical Books 123 Section Assessment Answers Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 123 Section Assessment Answers Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 123 Section Assessment Answers Chemistry
 - Setting Reading Goals 123 Section Assessment Answers Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 123 Section Assessment Answers Chemistry
 - Fact-Checking eBook Content of 123 Section Assessment Answers Chemistry

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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