

## 1992B. Answer the following two Periodic Table questions.

### Classifying the Periodic Table Grid

By using a color-coded periodic table, answer the next two questions.

#### Part (a)

In addition, the periodic table has been divided into groups. The organization of the table appears to be according to the type of ions an element will form. Identify the elements in the shaded area. Then use a periodic table to determine which group each element belongs to. How is the periodic table organized by properties, both chemical and physical? Illustrate your organization on the table below, either with the names of the elements or with symbols if known. Give the group for each element. The periodic table is organized into columns and rows. Elements in the same column have similar properties and elements in the same row have similar properties.

#### Blank Periodic Table


#### Part (b) Classify the elements in the shaded areas (show groups for each element)

Element 1: He, Ar, Kr, Xe, Rn, At

Element 2: Fr, Cs, Rb, K, Na, Li

Element 3: Hg, Au, Cu, Ag, Pt

Element 4: Zn, Cd, Sn, Pb

Element 5: Ba, Sr, Ca, Mg, Be

Element 6: P, As, Sb, Bi

Element 7: N, P, As, Sb, Bi

Element 8: Bromine: Br, I, At, Rn, Xe, Kr, Ar, Ne, He

Element 9: Fluorine: F, Cl, Br, I, At, Rn, Xe, Kr, Ar, Ne, He

# Pogil Cracking The Periodic Table Code Answers

**SJ Ball**



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