

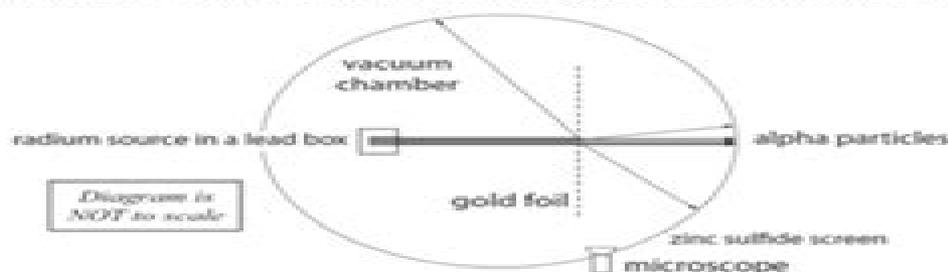
ATOMS: ATOMIC STRUCTURE QUESTIONS AND ANSWERS

QUESTION ONE: MODELS OF THE ATOM (2011:1)

At different times scientists have proposed various descriptions or models of the atom to match experimental evidence available.

- (a) The model that Thomson proposed was called the plum-pudding model. Describe this model.
Thomson's model had a number of negative particles / corpuscles / electrons embedded in a (solid) positive sphere / cloud.
Cloud of particles or energy not accepted.
The atom as a whole is neutral because the charges are balanced.
Term "nucleus" not accepted.

Geiger and Marsden performed a series of experiments under the direction of Ernest Rutherford which led to a new model of the atom. A model of the gold foil experiment is shown below.



- (b) For each of the conclusions given below, state which observation from the experiment provides evidence that:
- (i) Most of the mass of the atom is concentrated in a tiny region which Rutherford called the nucleus.
Most of the alpha particles went straight through and a very small number / a few bounced back / were reflected.
- (ii) The nucleus is positively charged.
Some of the alpha particles were deflected ("deflection" is an observation, "repulsion" is not).
- (c) Rutherford included several features in the experiment due to the characteristics of the alpha particles. Discuss the reason for each of the features below:
- (i) The source of the alpha particles is at the end of a narrow lead tube
Tube: To provide a narrow (or collimated) beam of alpha particle.
OR to direct the alpha particles
OR ensure alpha hits perpendicular to gold foil
OR to avoid stray alpha particles affecting the results
- (ii) Vacuum chamber
Vacuum chamber – Alpha particles would not travel far / be stopped / lose energy / path would change / be deviated / collide with the air particles (cannot accept "affect", "disrupt", "react", "interfere").
OR
Alpha would become a helium atom / alpha particles ionise the air.
- (iii) Zinc sulfide screen
Alpha particles are not visible to the naked eye. The ZnS coating will show / glow / fluoresce / flash / give off light when alpha particles strike it (this may be implied as opposed to directly stated) but do not allow "detected" or "located"

Atoms Atomic Structure Questions And Answers

LL Leslie



Atoms Atomic Structure Questions And Answers:

Decoding **Atoms Atomic Structure Questions And Answers**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Atoms Atomic Structure Questions And Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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Table of Contents Atoms Atomic Structure Questions And Answers

1. Understanding the eBook Atoms Atomic Structure Questions And Answers
 - The Rise of Digital Reading Atoms Atomic Structure Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Atoms Atomic Structure Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atoms Atomic Structure Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atoms Atomic Structure Questions And Answers
 - Personalized Recommendations
 - Atoms Atomic Structure Questions And Answers User Reviews and Ratings

- Atoms Atomic Structure Questions And Answers and Bestseller Lists
- 5. Accessing Atoms Atomic Structure Questions And Answers Free and Paid eBooks
 - Atoms Atomic Structure Questions And Answers Public Domain eBooks
 - Atoms Atomic Structure Questions And Answers eBook Subscription Services
 - Atoms Atomic Structure Questions And Answers Budget-Friendly Options
- 6. Navigating Atoms Atomic Structure Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Atoms Atomic Structure Questions And Answers Compatibility with Devices
 - Atoms Atomic Structure Questions And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atoms Atomic Structure Questions And Answers
 - Highlighting and Note-Taking Atoms Atomic Structure Questions And Answers
 - Interactive Elements Atoms Atomic Structure Questions And Answers
- 8. Staying Engaged with Atoms Atomic Structure Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atoms Atomic Structure Questions And Answers
- 9. Balancing eBooks and Physical Books Atoms Atomic Structure Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atoms Atomic Structure Questions And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atoms Atomic Structure Questions And Answers
 - Setting Reading Goals Atoms Atomic Structure Questions And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atoms Atomic Structure Questions And Answers
 - Fact-Checking eBook Content of Atoms Atomic Structure Questions And Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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