

Assignment #1 – Compound Names and Formulas
Single-valent ions only

A. Name these Compounds

- Li_2S Lithium Sulfide
- CaO Calcium Oxide
- NaF Sodium Fluoride
- CaBr_2 Calcium Bromide
- MgCl_2 Magnesium Chloride
- BBr_3 Boron Bromide
- Cs_2O Cesium Oxide
- FrBr Francium Bromide
- Ag_2S Silver Sulfide

- GeF_4 Germanium Fluoride
- Ga_2O_3 Gallium Oxide
- EsCl_3 Einsteinium Chloride
- Fm_2O_3 Fermium Oxide
- Mg_3N_2 Magnesium Nitride
- Rb_2O Rubidium Oxide
- RaO Radium Oxide
- SrO Strontium Oxide
- Tc_2O_7 Technetium Oxide

B. Write the correct chemical formula for these compounds by balancing the ionic charges

- sodium chloride NaCl
- magnesium fluoride MgF_2
- silver oxide Ag_2O
- indium bromide InBr_3
- zinc bromide ZnBr_2
- neodymium oxide Nd_2O_3
- thorium sulphide ThS_2
- actinium oxide Ac_2O_3
- radium bromide RaBr_2
- cesium oxide Cs_2O

- hydrogen oxide H_2O
- francium nitride Fr_3N
- rubidium phosphide Rb_3P
- potassium oxide K_2O
- beryllium sulphide BeS
- lithium sulphide Li_2S
- hydrogen bromide HBr
- strontium nitride Sr_3N_2
- calcium oxide CaO
- tantalum nitride Ta_3N_5

Writing And Naming Binary Compounds Worksheet Answer Key

Patrick Vollmar



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