

UNIT 10: PERIODIC TABLE **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

Section 1.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 16. A column in the periodic table | _____ a. metals |
| _____ 17. A row in the periodic table | _____ b. group |
| _____ 18. Group 1 elements | _____ c. period |
| _____ 19. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 20. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of platinum.
- _____ 22. Most elements besides boron of the elements in the periodic table are metalloids.
- _____ 23. Group 1A elements except for hydrogen are located on the left side.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are (a) nonmetals known as noble gases.
- _____ 27. Metalloids have properties of both metals and nonmetals.

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RJ Alexander



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Excel Science Study Guide, Years 9-10 Will Marchment, 2004 The book contains coverage of five major topic areas in the NSW School Certificate test Energy Force and Motion Atoms Elements and Compounds Structure and Function of Living Things Earth and Space Ecosystems Resources and Technology a chapter on Investigations and Problem Solving in Science to help with practical skills revision questions and chapter tests to help you remember important information a glossary and summary in each section of the book diagrams and illustrations to help your understanding a section to help you prepare for the School Certificate test a sample School Certificate test paper with answers answers to all questions **Chemistry**

McGraw-Hill Staff, 2001-07 For All Practical Purposes Consortium for Mathematics and Its Applications (U.S.), 2003 The sixth edition of the acclaimed classroom favorite offer a number of new features to help instructors strengthen the mathematical literacy of their students *Organic Chemistry* Francis A. Carey, 2000 A Market Leading Traditional Approach to Organic Chemistry Throughout all seven editions Organic Chemistry has been designed to meet the needs of the mainstream two semester undergraduate organic chemistry course This best selling text gives students a solid understanding of organic chemistry by stressing how fundamental reaction mechanisms function and reactions occur With the addition of handwritten solutions new cutting edge molecular illustrations updated spectroscopy coverage seamless integration of molecular modeling exercises and state of the art multimedia tools the 7th edition of Organic Chemistry clearly offers the most up to date approach to the study of organic chemistry **The Go-To Guide for Engineering Curricula, Grades**

9-12 Cary I. Sneider, 2014-12-05 How to engineer change in your high school science classroom With the Next Generation Science Standards your students won't just be scientists they'll be engineers But you don't need to reinvent the wheel Seamlessly weave engineering and technology concepts into your high school math and science lessons with this collection of time tested engineering curricula for science classrooms Features include A handy table that leads you straight to the chapters you need In depth commentaries and illustrative examples A vivid picture of each curriculum its learning goals and how it addresses the NGSS More information on the integration of engineering and technology into high school science education Chemistry Steven S. Zumdahl, Susan A. Zumdahl, 2000 Zumdahl's conceptual approach to problem solving early and thorough treatment of reactions and integration of descriptive chemistry and chemical principles throughout are key elements to the success of this bestseller By encouraging students to understand qualitative concepts before considering quantitative problems Chemistry helps students build a solid understanding of topics and avoid common misconceptions The text also includes new Chemical Impact boxes that present everyday applications of chemistry plus over 200 redesigned illustrations with greater clarity and emphasis on molecular level illustrations **DAT** Aftab S. Hassan, 1999 At Lippincott

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