

CHEMFILE MINI-GUIDE TO PROBLEM SOLVING**General Plan for Solving Dilution Problems**

$$1 \quad M_1V_1 = M_2V_2$$

Rearrange the equation $M_1V_1 = M_2V_2$ algebraically to solve for the unknown quantity.

$$2 \quad \begin{array}{c} \text{The equation used to} \\ \text{calculate the unknown quantity} \\ \text{will be one of the following four:} \\ V_2 = \frac{M_1V_1}{M_2}, \quad M_2 = \frac{M_1V_1}{V_2}, \quad V_1 = \frac{M_2V_2}{M_1}, \quad M_1 = \frac{M_2V_2}{V_1} \end{array}$$

Substitute each of the known quantities for its symbol, and calculate.

$$3 \quad \begin{array}{c} \text{Unknown} \\ \text{molarity or volume} \end{array}$$

SAMPLE PROBLEM 1

What is the molarity of a solution that is made by diluting 50.00 mL of a 4.74 M solution of HCl to 250.00 mL?

SOLUTION**1. ANALYZE**

- *What is given in the problem?* the molarity of the stock solution, the volume used to dilute, and the volume of the diluted solution
- *What are you asked to find?* the molarity of the diluted solution

Items	Data
Concentration of the stock solution (M_1)	4.74 M HCl
Volume of stock solution used (V_1)	50.00 mL
Volume of diluted solution (V_2)	250.00 mL
Concentration of the diluted solution (M_2)	? M

Chemfile Mini Guide To Problem Solving Answer Key

Xiang Xie



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Holt Chemistry File ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

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Scientific and Technical Aerospace Reports ,1977 **Holt ChemFile Lab Program** ,2005 [The Answer Key: A](#)

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