

Physics

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(Chapter 11) (Thermal Properties of Matter) (Exercises)

(Class 11)

Question 11.15:

Given below are observations on molar specific heats at room temperature of some common gases.

Gas	Molar specific heat (C_v) ($\text{cal mol}^{-1} \text{K}^{-1}$)
Hydrogen	4.87
Nitrogen	4.97
Oxygen	5.02
Nitric oxide	4.99
Carbon monoxide	5.01
Chlorine	6.17

The measured molar specific heats of these gases are markedly different from those for monatomic gases. Typically, molar specific heat of a monatomic gas is 2.92 cal/mol K . Explain this difference. What can you infer from the somewhat larger (than the rest) value for chlorine?

Answer 11.15:

The gases listed in the given table are diatomic. Besides the translational degree of freedom, they have other degrees of freedom (modes of motion).

Heat must be supplied to increase the temperature of these gases. This increases the average energy of all the modes of motion. Hence, the molar specific heat of diatomic gases is more than that of monatomic gases.

If only rotational mode of motion is considered, then the molar specific heat of a diatomic gas

$$= \frac{5}{2}R \Rightarrow \frac{5}{2} \times 1.98 = 4.95 \text{ cal mol}^{-1} \text{K}^{-1}$$

With the exception of chlorine, all the observations in the given table agree with $\left(\frac{5}{2}R\right)$.

This is because at room temperature, chlorine also has vibrational modes of motion besides rotational and translational modes of motion.

Question 11.16:

A child running a temperature of 101°F is given an antipyrin (i.e. a medicine that lowers fever) which causes an increase in the rate of evaporation of sweat from his body. If the fever is brought down to 98°F in 20 minutes, what is the average rate of extra evaporation caused, by the drug. Assume the evaporation mechanism to be the only way by which heat is lost. The mass of the child is 30 kg. The specific heat of human body is approximately the same as that of water, and latent heat of evaporation of water at that temperature is about 580 cal g^{-1} .

Answer 11.16:

Initial temperature of the body of the child, $T_1 = 101^\circ\text{F}$

Final temperature of the body of the child, $T_2 = 98^\circ\text{F}$

$$\text{Change in temperature, } \Delta T = \left[(101 - 98) \times \frac{5}{9} \right]^\circ\text{C}$$

Time taken to reduce the temperature, $t = 20 \text{ min}$

Mass of the child, $m = 30 \text{ kg} = 30 \times 10^3 \text{ g}$

Specific heat of the human body = Specific heat of water = $c = 1000 \text{ cal/kg/}^\circ\text{C}$

Latent heat of evaporation of water, $L = 580 \text{ cal g}^{-1}$

The heat lost by the child is given as: $\Delta\theta = mc\Delta T = 30 \times 1000 \times (101 - 98) \times \frac{5}{9} = 50000 \text{ cal}$

Physics Study Guide Answers Chapter 11

Bo Lou



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Study Guide and Student Solutions Manual for Wilson College Physics David Cordes, Jerry D. Wilson, John Kinard, 1990 This reader friendly book presents the fundamental principles of physics in a clear and concise manner Emphasizing conceptual understanding as the basis for mastering a variety of problem solving tools it provides a wide range of relevant applications and illustrative examples This book discusses mechanics thermodynamics and oscillations and wave motion For anyone wishing to learn more about the fundamentals of physics and how physical principles apply to a variety of real world situations devices and topics *Student Study Guide & Selected Solutions Manual* Frank L. H Wolfs, 2008

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Physics, , Student Study Guide John D. Cutnell, Kenneth W. Johnson, 1997-09-04 Describes applications in medicine automobile features transportation home entertainment athletics household applications information processing detection devices camera technology and many more Contains numerous discussions and examples that focus on human physiology including muscle forces blood pressure the refraction of light by the eye and many others **Student Study Guide to Accompany Fundamentals of Physics, Rev. Printing, Physics, 3d Ed., Parts 1 and 2** Stanley A. Williams, 1978

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carefully constructed to be of maximum use to the students With few exceptions we adhere to the following policies The section exercises are graded into three consecutive groups a The first exercises are routine modelled almost exactly on the exam ples these are intended to give students confidence b Next come exercises that are still based directly on the examples and text but which may have variations of wording or which combine different ideas these are intended to train students to think for themselves c The last exercises in each set are difficult These are marked with a star and some will challenge even the best students Difficult does not necessarily mean theoretical often a starred problem is an interesting application that requires insight into what calculus is really about The exercises come in groups of two and often four similar ones

Calculus II Jerrold Marsden, Alan Weinstein, 2012-12-06 The second of a three volume work this is the result of the authors experience teaching calculus at Berkeley The book covers techniques and applications of integration infinite series and differential equations the whole time motivating the study of calculus using its applications The authors include numerous solved problems as well as extensive exercises at the end of each section In addition a separate student guide has been prepared A+ Complete Study Guide David Groth, Dan Newland, 2002 In this proven Sybex study guide in depth coverage of every exam objective is provided The book includes practical information on essential hardware and OS related tasks with hundreds of challenging review questions in the book and on the CD **Project Physics Course: Text and Handbook: The triumph of mechanics** Harvard Project Physics, 1970 **Introductory College Physics** Joseph F. Mulligan, David A. Jerde, 1990-12

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