

## Chemistry – Unit 10 Worksheet 3

1. Element X has two natural isotopes. The isotope with a mass number of 6 has a relative abundance of 7.5%. The isotope with a mass number of 7 has a relative abundance of 92.5%. Determine the average molar mass for the element from these figures. What is the true identity and atomic number of element X?
2. The element copper is found to contain the naturally occurring isotopes  ${}^{63}\text{Cu}$  and  ${}^{65}\text{Cu}$ . The relative abundances are 69.1% and 30.9% respectively. Calculate the average molar mass of copper.
3. Uranium has three isotopes with the following percent abundances:  ${}^{234}\text{U}$  (0.0058%),  ${}^{235}\text{U}$  (0.71%),  ${}^{238}\text{U}$  (99.23%). What do you expect the molar mass of uranium to be in whole numbers? Why?
4. A sample of silver as it occurs in nature is 52.0% of isotope  ${}^{107}\text{Ag}$  and 48.0% of isotope  ${}^{109}\text{Ag}$ . What is the average molar mass of silver? (Compare your result with the value given in the periodic table).

# Modeling Chemistry U9 Ws 2 V2 Answers

**Daniel F McAuley**



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## Adopting the Track of Phrase: An Emotional Symphony within **Modeling Chemistry U9 Ws 2 V2 Answers**

In a global used by screens and the ceaseless chatter of instantaneous connection, the melodic beauty and mental symphony produced by the prepared word usually disappear into the back ground, eclipsed by the relentless noise and distractions that permeate our lives. However, nestled within the pages of **Modeling Chemistry U9 Ws 2 V2 Answers** a marvelous fictional treasure full of raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, that fascinating masterpiece conducts viewers on an emotional trip, well unraveling the concealed tunes and profound influence resonating within each cautiously crafted phrase. Within the depths of the poignant evaluation, we shall discover the book is central harmonies, analyze its enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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