

Chemistry Worksheet Class 10 on Chapter 5 Periodic Classification of Elements - Set 3

Q1. How many elements are there in the fifth period of the periodic table?

- (a) Eighteen
- (b) Thirty-two
- (c) Eight
- (d) None of the above

Q2. Fourteen elements after lanthanum are known as

- (a) Lanthanides
- (b) Actinides
- (c) d- block elements
- (d) None of the above

Q3. What is the electronic configuration of Al^{3+} ion?

- (a) 2, 8
- (b) 2, 8, 1
- (c) 2, 8, 2
- (d) 2, 8, 3

Q4. Which group elements are also known as transition metals?

- (a) s block elements
- (b) p block elements
- (c) d block elements
- (d) f block elements

Q5. A metal belongs to the first group of the modern periodic table. What is the formula of its oxide?

- (a) MO
- (b) MO_2
- (c) M_2O
- (d) None of the above

Q6. What is the basis for classifying elements of the Mendeleev periodic table?

Q7. Let X be an element with an Atomic number equivalent to nineteen. What is the formula of its chloride?

Q8. The atomic number of elements U, V, X, Y and Z are given below.

Element	U	V	X	Y	Z
Atomic Number	7	10	12	4	19

(i) Which element belongs to the 3rd period of the periodic table?

(ii) Which element among these is a non-metal?

Q9. An element belongs to group 17. It is present in the third period, and its atomic number is 17. What is the element's atomic number belonging to the same group and present in the fifth period?

Chapter 5 The Periodic Table Answers

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