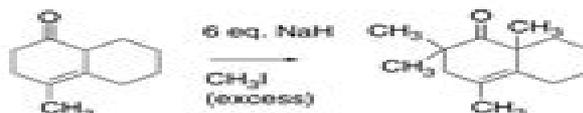


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through γ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH₂ in liquid NH₃. The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH₂ in liquid NH₃.



- e. The reaction was carried out using 1,1,1,3,3,3-hexafluoroisopropanol as the solvent. This is a very good solvent for the reaction and is also a very good solvent for the product.

- f. The reaction was carried out using 1,1,1,3,3,3-hexafluoroisopropanol as the solvent. This is a very good solvent for the reaction and is also a very good solvent for the product. The reaction was carried out using 1,1,1,3,3,3-hexafluoroisopropanol as the solvent. This is a very good solvent for the reaction and is also a very good solvent for the product.



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