

Acid-Base Titration Problems

Find the requested quantities in the following problems:

- 1 If it takes 54 mL of 0.1 M NaOH to neutralize 125 mL of an HCl solution, what is the concentration of the HCl?
- 2 If it takes 25 mL of 0.05 M HCl to neutralize 345 mL of NaOH solution, what is the concentration of the NaOH solution?
- 3 If it takes 15.0 mL of 0.40 M NaOH to neutralize 5.0 mL of HCl, what is the molar concentration of the HCl solution?
- 4 It takes 38 mL of 0.75 M NaOH solution to completely neutralize 155 mL of a sulfuric acid solution (H_2SO_4). What is the concentration of the H_2SO_4 solution?
- 5 If it takes 10.0 mL of 2.0 M H_2SO_4 to neutralize 30.0 mL of KOH, what is the molar concentration of the KOH?

Acid Base Titration Practice Problems With Answers

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