

# MOLE TO MOLE STOICHIOMETRY WORKSHEET

Name : \_\_\_\_\_

Date : \_\_\_\_\_

1. Using the equation



a. Find the number of moles of Al used to produce 9.9 moles of  $\text{H}_2$ .

b. Find the number of moles of HCl used to produce 1.11 moles of  $\text{AlCl}_3$ .

c. Find the number of moles of  $\text{AlCl}_3$  produced along with 0.0757 moles of  $\text{H}_2$ .

2. The reaction between lithium and fluorine produces lithium fluoride. How many moles of lithium fluoride can be produced with 2.17 mol of fluorine?



3. In the reaction of copper (II) oxide breaking down to form copper and oxygen, how much copper is produced if 0.663 moles of oxygen is given off?



4. Fermentation is a complex chemical process of making wine by converting glucose into ethanol and carbon dioxide:



If 17.5 moles of ethanol were produced, how many moles of glucose were there initially?

5. Consider the reaction:



Calculate the moles of HCl needed to react completely with 8.25 moles of zinc.

# Moles And Stoichiometry Practice Problems Answers

**Pamela Mills, Amina El-Ashmawy**



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