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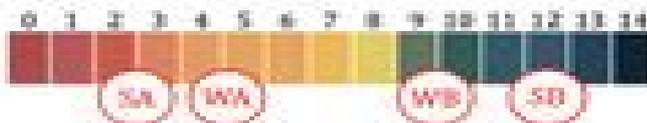
Acid-Base Solutions – PhET Simulator

Instructions: Open the Acid-Base Solutions simulator on the PhET website/app. Choose "Introduction."

- **pH Test Strips** change color when they come in contact with a solution.
- The color of the strip is compared to a **pH Color Key** to determine the pH.



- 1) Select the **pH Test Strip** from the **Tools** menu. Dip it into water. What is the pH? 7
- 2) Use pH test strips to test the **Strong Acid**, **Weak Acid**, **Strong Base**, and **Weak Base**. Draw the symbol for each solution in the appropriate location on the **pH Color Key** below:



Draw each symbol under the correct pH

SA	WA	SB	WB
Strong Acid	Weak Acid	Strong Base	Weak Base

- When acids and bases dissolve in water, their ions **dissociate** (separate).
- Acids dissociate by releasing **H⁺ ions**. These **H⁺ ions** bond to **H₂O** to form **H₃O⁺**.

- 3) Choose the **Strong Acid**. Complete the chemical formula for dissociation of a strong acid:



Instructions: Under "Views," choose the **Graph** option.

- 4) What is the concentration of **ions** (A⁻ and H₃O⁺) in the **Strong Acid**? 1.00×10^{-2}
- 5) What is the concentration of **ions** (A⁻ and H₃O⁺) in the **Weak Acid**? 3.16×10^{-5}
- 6) Which type of acid (**Strong** or **Weak**) produces more **ions**? **Strong Acid**



DIGITAL LAB

Acid And Bases Ph Phet Lab Answers

JA Banks



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