

## Balancing Chemical Equations

Once we have written a chemical equation, we must make sure to balance it! Balancing a chemical equation means to make sure that there are the same number of each type of atom on both sides of the chemical equation. So, if there are five carbons on the left side (reactants), there needs to be five carbons on the right side (products).

So, how do we balance an equation? Follow the following steps...

- Change **only** stoichiometric coefficients – those big numbers in front of the compounds (not subscripts in the formulas – that would change the identity of the compound!)
- Start with most complex
- If found in more than one place on each side, do it last
- Watch out for odd numbers (make them into even ones if needed)
- Check your work!

Now, try to balance the following equations. In addition, name the bolded compounds for some nomenclature practice!

- $\underline{\quad} \text{NH}_4\text{NO}_3(\text{s}) \rightarrow \underline{\quad} \text{N}_2\text{O}(\text{g}) + \underline{2} \text{H}_2\text{O}(\text{g})$   
*ammonium nitrate & dinitrogen monoxide*
- $\underline{\quad} \text{CH}_3\text{CO}_2\text{H}(\text{aq}) + \underline{2} \text{O}_2(\text{g}) \rightarrow \underline{2} \text{CO}_2(\text{g}) + \underline{2} \text{H}_2\text{O}(\text{l})$   
*oxygen & carbon dioxide*
- $\underline{2} \text{Pb}(\text{NO}_3)_2(\text{s}) \rightarrow \underline{2} \text{N}_2\text{O}_4(\text{g}) + \underline{2} \text{PbO}(\text{s}) + \underline{\quad} \text{O}_2(\text{g})$   
*lead(II) nitrate & dinitrogen tetroxide & lead(II) oxide*
- $\underline{\quad} \text{SiO}_2(\text{s}) + \underline{4} \text{HF} \rightarrow \underline{\quad} \text{SiF}_4(\text{g}) + \underline{2} \text{H}_2\text{O}(\text{l})$   
*silicon dioxide & silicon tetrafluoride*
- $\underline{8} \text{H}_2\text{S}(\text{aq}) + \underline{8} \text{Cl}_2(\text{g}) \rightarrow \underline{\quad} \text{S}_8(\text{s}) + \underline{16} \text{HCl}(\text{aq})$   
*chlorine*
- $\underline{\quad} \text{H}_2\text{CO}_3(\text{aq}) + \underline{\quad} \text{Ba}(\text{OH})_2 \rightarrow \underline{\quad} \text{BaCO}_3(\text{s}) + \underline{2} \text{H}_2\text{O}(\text{l})$   
*barium hydroxide & barium carbonate*
- $\underline{\quad} \text{P}_4(\text{s}) + \underline{6} \text{Br}_2(\text{l}) \rightarrow \underline{4} \text{PBr}_3(\text{l})$   
*phosphorous & phosphorous tribromide*
- $\underline{\quad} \text{S}_8(\text{l}) + \underline{4} \text{Cl}_2(\text{g}) \rightarrow \underline{4} \text{S}_2\text{Cl}_2(\text{g})$   
*disulfur dichloride*
- $\underline{2} \text{B}_5\text{H}_9(\text{l}) + \underline{12} \text{O}_2(\text{g}) \rightarrow \underline{5} \text{B}_2\text{O}_3(\text{s}) + \underline{9} \text{H}_2\text{O}(\text{g})$   
*pentaboron nonahydride & diboron trioxide*
- $\underline{3} \text{H}_2\text{SO}_4(\text{aq}) + \underline{2} \text{Al}(\text{OH})_3(\text{s}) \rightarrow \underline{\quad} \text{Al}_2(\text{SO}_4)_3(\text{aq}) + \underline{6} \text{H}_2\text{O}(\text{l})$   
*aluminum hydroxide & aluminum sulfate*
- $\underline{2} \text{Al}(\text{s}) + \underline{3} \text{Br}_2(\text{l}) \rightarrow \underline{2} \text{AlBr}_3(\text{s})$   
*aluminum bromide*
- $\underline{\quad} (\text{NH}_4)_2\text{Cr}_2\text{O}_7(\text{s}) \rightarrow \underline{\quad} \text{N}_2(\text{g}) + \underline{\quad} \text{Cr}_2\text{O}_3(\text{s}) + \underline{4} \text{H}_2\text{O}(\text{l})$   
*ammonium dichromate & chromium(III) oxide*
- $\underline{\quad} \text{S}_8(\text{s}) + \underline{24} \text{F}_2(\text{g}) \rightarrow \underline{8} \text{SF}_6(\text{g})$   
*sulfur hexafluoride*
- $\underline{\quad} \text{Zn}(\text{s}) + \underline{\quad} \text{Pb}(\text{NO}_3)_2(\text{aq}) \rightarrow \underline{\quad} \text{Zn}(\text{NO}_3)_2(\text{aq}) + \underline{\quad} \text{Pb}(\text{s})$   
*zinc nitrate*

# Balancing Chemical Equations Multiple Choice Test

**Jiyuan Zhang**



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web oct 1 2023 aisi 1040 carbon steel has high carbon content and can be hardened by heat treatment followed by quenching and tempering to achieve 150 to 250 ksi tensile strength chemical composition the following table shows the chemical composition of aisi 1040 carbon steel element content iron fe 98 6 99 manganese mn 0 60

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web carbon and alloy steel grades sae designation type and composition by weight carbon steels 10xx plain carbon mn 1 00 max 11xx resulfurized 12xx resulfurized and rephosphorized 15xx plain carbon mn 1 00 1 65 max manganese steels 13xx mn 1 75 nickel steels 23xx ni 3 50 25xx ni 5 00 nickel chromium steels 31xx ni

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web jul 10 2012 from wikipedia the free encyclopedia the society of automotive engineers sae designates sae steel grades these are four digit numbers which represent chemical composition standards for steel specifications the american iron and steel institute aisi originally started a very similar system

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web high carbon steel high carbon steel has a carbon content of 0 60 1 25 wt and a manganese content of 0 30 0 90 wt it has the highest hardness and toughness of the carbon steels and the lowest ductility high carbon steels are very wear resistant as a result of the fact that they are almost always hardened and tempered

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web the chemical composition of aisi 1215 carbon steel is tabulated below physical properties the following table shows the physical properties of aisi 1215 carbon steel mechanical properties the mechanical properties of the cold drawn aisi 1215 carbon steel are given in the following table thermal properties

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web dec 7 2009 chemical compositions of sae carbon steels j403 200912 in 1941 the sae iron and steel division in collaboration with the american iron and steel institute aisi made a major change in the method of expressing composition ranges for the sae steels the plan as now applied is based in general on narrower cast or heat analysis

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web a587 22 standard specification for electric resistance welded low carbon steel pipe for the chemical industry structural quality furnished to chemical composition requirements a633 a633m 18 standard specification for normalized high strength low alloy structural steel plates a827 a827m 14 2018 standard specification for plates

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web carbon steel is a steel with carbon content from about 0 05 up to 2 1 percent by weight the definition of carbon steel from the american iron and steel institute aisi states no minimum content is specified or required for chromium cobalt

molybdenum nickel niobium titanium tungsten vanadium zirconium or any other element to be added to obtain a  
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web below is material aisi 1018 carbon steel properties and specifications including chemical composition physical and mechanical properties such as density thermal expansion coefficient thermal conductivity tensile yield strength and hardness etc aisi sae 1018 chemical composition

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web mit department of civil and environmental engineering spring semester 1999 chemical composition of structural steels the primary types of structural steel are usually classified according to the following chemical composition categories carbon manganese steels high strength low alloy hsla steels