

- 6 (a) i. Using chemical equations only, distinguish between ethene and ethyne. (3 marks)
- ii. Name the compound formed when ethyne polymerises. (1 mark)
- iii. Write the structure of the compound in a(ii). (1 mark)
- (b) i. What will be observed when:
- (I) Fehling's solution is added to reducing sugar?
  - (II) Millon's reagent is added to protein?
  - (III) Sodium metal is dropped into a beaker containing ethanol? (3 marks)
- (ii) Calculate the solubility in  $\text{mol/dm}^3$  of 4.9 g  $\text{KClO}_3$  in  $20 \text{ cm}^3$  of water at  $66^\circ\text{C}$ .  
[K = 39, Cl = 35.5, O = 16] (5 marks)
- (c) i. State three physical properties of ethanol. (3 marks)
- ii. Write the equations for the preparation of ethanol from ethene. (4 marks)
- (d) i. Mention two uses of sodium trioxocarbonate (IV). (2 marks)
- ii. State three properties of transition elements. (3 marks)

# Chemistry Question And Answer Theory Objectives

**William Emerson Ritter**



## **Chemistry Question And Answer Theory Objectives:**

## The Enigmatic Realm of **Chemistry Question And Answer Theory Objectives**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Chemistry Question And Answer Theory Objectives** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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