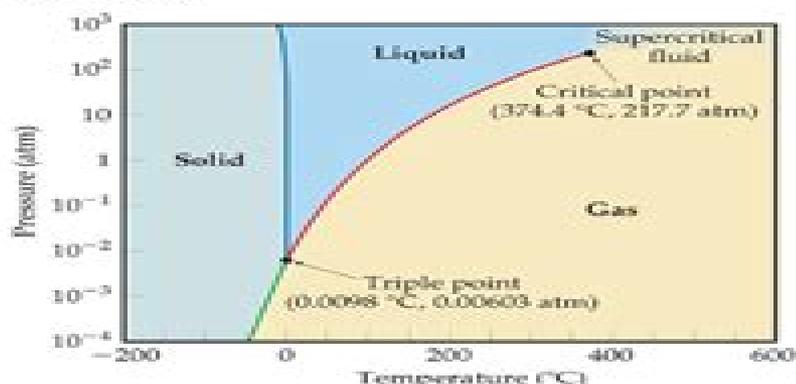


CHEMISTRY 121

PART 1 #1-10 are true/false (20 points)

1. ____ The internal energy of a system is increased by having the surrounding do work on the system
2. ____ Colligative properties do not depend on the concentration of the solute
3. ____ When dissolved in water, bases increase the concentration of OH⁻ ions
4. ____ Reducing agent (reducer) accepts electrons in the redox reaction
5. ____ Oxidation number of C in CO₃²⁻ is +2
6. ____ Reaction between an acid and a base is a redox reaction
7. ____ To obtain 0.25 M solution of sucrose, you have to mix 1 part of 1 M sucrose solution and 4 parts of water
8. ____ Tin of baked beans is an isolated thermodynamic system
9. ____ All acids only partially dissociate into constituent ions when dissolved
10. ____ At a pressure of 0.001 atm and temperature of 50°C water will be in a gaseous state(vapor)



PART 2. Each question is worth 5 points.

11. Which two compounds are strong electrolytes:
 - a. NaOH and HBr

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DJ Losen



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