Name _____ Class _____ Date _____

Advanced Level

Chapter Test B

Chapter: Chemical Equations and Reactions

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. The precipitation of a slightly soluble solid compound by a double-displacement reaction results in the formation of a
a. gas.
b. precipitate.
c. combination reaction.
d. balance.

_____ 2. To balance a chemical equation, it is permissible to adjust the
a. coefficients.
b. subscripts.
c. formulas of the products.
d. number of participants.

_____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
a. solid.
b. dissolved in water.
c. an acid.
d. insoluble.

_____ 4. The tendency for a replacement reaction to occur depends on the
a. relative tendency for each element to lose electrons in the activity series.
b. temperature of the reaction.
c. relative electronegativity.
d. relative tendency for each element to gain electrons in the activity series.

_____ 5. The coefficients in a chemical equation
a. indicate the number of moles of each substance that react.
b. show the number of atoms of each substance that react.
c. are the molar masses of the substances.
d. show the relative abundance of each atom.

_____ 6. If one of X in Brown State provides X in the activity series, then
a. X will replace some of Y in solution.
b. Y will replace some of X in solution.
c. X will cause cooling only in Brown.
d. X will react with cold water.

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Chemistry Chapter 8 Study Guide Answers

Paul B. Kelter

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