

## 9

## CHEMICAL NAMES AND FORMULAS

## Practice Problems

In your notebook, solve the following problems.

## SECTION 9.1 NAMING IONS

- What is the charge on the ion typically formed by each element?
  - oxygen
  - iodine
  - sodium
  - aluminum
  - nickel, 2 electrons lost
  - magnesium
- How many electrons does the neutral atom gain or lose when each ion forms?
  - $\text{Cr}^{3+}$
  - $\text{P}^{3-}$
  - $\text{Li}^+$
  - $\text{Ca}^{2+}$
  - $\text{Cl}^-$
  - $\text{O}^{2-}$
- Name each ion. Identify each as a cation or an anion.
  - $\text{Sr}^{2+}$
  - $\text{Co}^{2+}$
  - $\text{Br}^-$
  - $\text{K}^+$
  - $\text{H}^+$
  - $\text{Mn}^{2+}$
- Write the formula (including charge) for each ion. Use Table 9.3 if necessary.
  - carbonate ion
  - nitrite ion
  - sulfate ion
  - hydroxide ion
  - chromate ion
  - ammonium ion
- Name the following ions. Identify each as a cation or an anion.
  - $\text{CN}^-$
  - $\text{HCO}_3^-$
  - $\text{PO}_4^{3-}$
  - $\text{Cl}^-$
  - $\text{Ca}^{2+}$
  - $\text{SO}_3^{2-}$

## SECTION 9.2 NAMING AND WRITING FORMULAS FOR IONIC COMPOUNDS

- Write the formulas for these binary ionic compounds.
  - magnesium oxide
  - tin(II) fluoride
  - potassium iodide
  - aluminum chloride
  - sodium sulfide
  - ferrous bromide
- Write the formulas for the compounds formed from these pairs of ions.
  - $\text{Ba}^{2+}$ ,  $\text{Cl}^-$
  - $\text{Ag}^+$ ,  $\text{I}^-$
  - $\text{Ca}^{2+}$ ,  $\text{S}^{2-}$
  - $\text{K}^+$ ,  $\text{Br}^-$
  - $\text{Al}^{3+}$ ,  $\text{O}^{2-}$
  - $\text{Fe}^{2+}$ ,  $\text{O}^{2-}$
- Name the following binary ionic compounds.
  - $\text{MnO}_2$
  - $\text{Li}_2\text{N}$
  - $\text{CaCl}_2$
  - $\text{SrBr}_2$
  - $\text{NiCl}_2$
  - $\text{K}_2\text{S}$
  - $\text{CuCl}_2$
  - $\text{SnCl}_4$

# Chapter 9 Chemical Names And Formulas Test B Answers

**Peter Atkins, Loretta Jones, Leroy  
Laverman**

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The Medical World ,1904      **General, Organic and Biochemistry** Kate Brown,William Henry Brown,1983

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## **Table of Contents Chapter 9 Chemical Names And Formulas Test B Answers**

1. Understanding the eBook Chapter 9 Chemical Names And Formulas Test B Answers
  - The Rise of Digital Reading Chapter 9 Chemical Names And Formulas Test B Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 9 Chemical Names And Formulas Test B Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 9 Chemical Names And Formulas Test B Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 9 Chemical Names And Formulas Test B Answers
  - Personalized Recommendations
  - Chapter 9 Chemical Names And Formulas Test B Answers User Reviews and Ratings
  - Chapter 9 Chemical Names And Formulas Test B Answers and Bestseller Lists
5. Accessing Chapter 9 Chemical Names And Formulas Test B Answers Free and Paid eBooks
  - Chapter 9 Chemical Names And Formulas Test B Answers Public Domain eBooks
  - Chapter 9 Chemical Names And Formulas Test B Answers eBook Subscription Services
  - Chapter 9 Chemical Names And Formulas Test B Answers Budget-Friendly Options

6. Navigating Chapter 9 Chemical Names And Formulas Test B Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 9 Chemical Names And Formulas Test B Answers Compatibility with Devices
  - Chapter 9 Chemical Names And Formulas Test B Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 9 Chemical Names And Formulas Test B Answers
  - Highlighting and Note-Taking Chapter 9 Chemical Names And Formulas Test B Answers
  - Interactive Elements Chapter 9 Chemical Names And Formulas Test B Answers
8. Staying Engaged with Chapter 9 Chemical Names And Formulas Test B Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 9 Chemical Names And Formulas Test B Answers
9. Balancing eBooks and Physical Books Chapter 9 Chemical Names And Formulas Test B Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 9 Chemical Names And Formulas Test B Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter 9 Chemical Names And Formulas Test B Answers
  - Setting Reading Goals Chapter 9 Chemical Names And Formulas Test B Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 9 Chemical Names And Formulas Test B Answers
  - Fact-Checking eBook Content of Chapter 9 Chemical Names And Formulas Test B Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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web although we didn t use the word retrospective in the title this exhibition and the extended interview in this catalog are intended to serve as a retrospective exploration of the strip and watterson s journey as an artist from his early efforts as a student until he ended calvin and hobbes in 1995

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