

## CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfide, the roman numeral tells us (2)
  - How many atoms of Fe are in one formula unit.
  - How many sulfide ions can be attached to the iron atom.
  - The charge on each Fe ion.
  - The total positive charge of the formula unit.
- The result of changing a subscript in a correctly written chemical formula is (1)
  - Change the number of ions represented in the formula.
  - Change the charges on the other ions in the compound.
  - Change the formula so that it no longer represents the compound.
  - Have no effect on the formula.
- Derive a generalization for when an acid name will end in the suffix -ic or -ous. If ending in -ic, the anion ended in -ate. If ending in -ous, the anion ended in -ite.
- Derive a generalization for when an acid name will begin with the prefix hydro-. If prefix is hydro-, the anion ended in -ite. This is almost always a monatomic ion.
- Fill in the blanks in the table below.

Compound Name	Formula
Acetic acid	$\text{CH}_3\text{CO}_2\text{H}$
Aluminum sulfate	$\text{Al}_2(\text{SO}_4)_3$
Lead II carbonate	$\text{PbCO}_3$
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Sulfuric acid	$\text{H}_2\text{SO}_4$
Nitrogen pentoxide	$\text{N}_2\text{O}_5$
Copper II carbonate	$\text{CuCO}_3$
Sodium sulfate	$\text{Na}_2\text{SO}_4$
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Mercury sulfate	$\text{HgSO}_4$
Nitrous acid	$\text{HNO}_2$
Magnesium perchlorate	$\text{Mg}(\text{ClO}_4)_2$
Iron II nitrate	$\text{Fe}(\text{NO}_3)_2$
Iron III nitrate	$\text{Fe}(\text{NO}_3)_3$
Carbon Dioxide	$\text{CO}_2$
Sulfur dioxide	$\text{SO}_2$
Sulfur pentachloride	$\text{SCl}_5$
Tetraphosphorus hexasulfide	$\text{P}_4\text{S}_6$
Tetrasulfur dioxide	$\text{S}_4\text{O}_2$

- Name the following compound and identify its molecular ions or acid.
  - $\text{Ca}(\text{NO}_3)_2$ , calcium (II) nitrate - II
  - $\text{Fe}(\text{NO}_3)_3$ , Iron (III) nitrate - III
  - $\text{Fe}(\text{OH})_2$ , Iron (II) hydroxide - II
  - $\text{N}_2\text{O}_5$ , Nitrogen pentoxide - III

# Chapter 9 Chemical Names And Formulas Quiz Answers

**Michael Hoelscher**



## **Chapter 9 Chemical Names And Formulas Quiz Answers:**

## Decoding **Chapter 9 Chemical Names And Formulas Quiz Answers**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chapter 9 Chemical Names And Formulas Quiz Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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