

**1E1004**

Roll No. \_\_\_\_\_

Total No of Pages: **3****1E1004**

**B. Tech I Sem. (Back) Exam. Jan. 2016**  
**203 Old Engineering Chemistry**  
**Common to all Branches of Engineering**

**Time: 3 Hours****Maximum Marks: 80**  
**Min. Passing Marks: 24***Instructions to Candidates:*

*Attempt overall five questions, selecting one question from each unit. All questions carry equal marks.*

*Use of following supporting material is permitted during examination. (Mentioned in form No. 205)*

1. NIL2. NIL**UNIT-I**

- Q.1 (a) What is EDTA? How will you determine the hardness of water by EDTA? [3+3=6]
- (b) What is Break point chlorination? [4]
- (c) Water containing  $\text{CaCO}_3$  120 mg/lit,  $\text{MgCO}_3$  84 mg/lit,  $\text{Mg}(\text{NO}_3)_2$  74 mg/lit,  $\text{CaCl}_2$  111 mg/lit and  $\text{KCl}$  100 mg/lit is to be treated with lime soda. Calculate the amount of chemicals required for the treatment of 5000 lit of water. [6]

**OR**

- Q.1 (a) Describe lime soda process for water softening. Give the chemical reactions involved during the softening. [3+3=6]
- (b) If silica is present in water, what harmful effects it can cause? [4]
- (c) Calculate the amount of lime required for softening of 5000 lit of hard water containing 72 ppm of  $\text{Mg SO}_4$ . [6]

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